KWAME NKRUMAH UNIVERSITY OF SCIENCE AND TECHNOLOGY,

## **KUMASI, GHANA**

## BIOLOGICAL ACTIVITIES OF *HILLERIA LATIFOLIA* (LAM.) H. WALT AND *LAPORTEA OVALIFOLIA* (SCHUMACH.) CHEW

By

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## DECLARATION

The experimental work described in this thesis was carried out at the Department of Pharmaceutics, Faculty of Pharmacy and Pharmaceutical Sciences, College of Health Science, KNUST, Kumasi, Ghana. I hereby declare that to the best of my knowledge this work has not been submitted for any other degree.

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## ABSTRACT

Plants have been accepted as part of human culture for the treatment of diseases and ailments, with about 80% of the populace in developing countries depending on plants for medicinal purposes. However, few plants have been scientifically proven for their folkloric uses and assessed for their quality, efficacy and toxicity. The study therefore sought to evaluate the antimicrobial, antioxidant, anti-inflammatory and wound healing activities as well as cytotoxic effects of *Hilleria latifolia* (Lam.) H. Walt.

(Family Phytolaccaceae) and Laportea ovalifolia (Schumach.) Chew. (Family Urticaceae). These plants are used locally for the treatment of wounds, general oedema and skin diseases. The broth dilution method was used to assess the microbial susceptibility and MIC against typed and clinical strains of Gram-positive, Gramnegative bacteria and fungi. The effect of sub-inhibitory concentration on some antibiotics was also determined. The antioxidant activity was investigated employing the DPPH free radical scavenging assay, the total phenolic content as well as the total antioxidant capacity assays. The carrageenan-induced oedema model in rat was used to evaluate the anti-inflammatory activity of the plants. The wound healing activity was established using the excision wound healing model in rats. The MTT-assay for cell viability and lactate dehydrogenase (LDH) assay were used to assess the cytotoxic effect of the plant extracts. The MICs of Hilleria latifolia leaf methanol extract (HLML), Hilleria latifolia root methanol extracts (HLMR) and Laportea ovalifolia leaf methanol extract (LOML) ranged from 50 to100 mg/mL against test microorganisms. The sub-inhibitory concentration (5 mg/mL) modified the activity of amoxicillin, erythromycin, ciprofloxacin, tetracycline and ampicillin against the test microorganisms by either potentiating or reducing their antimicrobial activity. HLML,

HLMR and LOML exhibited radical scavenging activity with IC<sub>50</sub> values of 102.5  $\pm$ 

1.5, 233.5  $\pm$  0.5 and 130.8  $\pm$  0.9 µg/mL. Total phenol content and total antioxidant capacity increased as concentration increased. The total phenol content of HLML, HLMR and LOML was calculated to be 103.0  $\pm$  1.335, 91.32  $\pm$  4.258 and 56.75  $\pm$ 

0.3220 mg/g, respectively. The total antioxidant capacity was  $410.4 \pm 4.732$  mg/g for HLML,  $408.0 \pm 18.70$  mg/g for HLMR and  $337.6 \pm 6.961$  mg/g for LOML. The antiinflammation activity of HLML, HLMR and LOML at 300 mg/kg caused a significant (p<0.001) decrease in oedema when administered before and after inducing oedema. There was also significant (p<0.001) increase in wound contraction from days 5 to15 after injury when treated with 5 and 10% HLML, HLMR and LOML, with increased fibroblast proliferation and collagenation as well as reepithelialisation. Treatment with 100 µg/mL of HLML (p<0.001) and HLMR

(p<0.05) significantly caused a reduction in cell viability, whiles cells treated with LOML showed no significant (p>0.05) difference in viability when compared to the untreated cells. HaCaT-keratinocytes treated with HLML, HLMR and LOML (0.1, 1.0, 10.0 and 100.0 µg/mL) exhibited no significant (p>0.05) LDH release when compared with untreated cells. The study therefore indicated that HLML, HLMR and LOML possess antimicrobial, antioxidant, anti-inflammatory and wound healing activities and they did not exhibit cytotoxic activities at the concentrations used.

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## LIST OF PUBLICATIONS

- Dapaah, S. O., Agyare, C., Kessieh, E. and Boakye, Y. D. (2015). Wound healing and cytotoxicity studies of methanol extracts of *Hilleria latifolia* and *Laportea ovalifolia*. In: Ghana Biomedical Convention, conference 2015, 707: p 86.
- Dapaah, S. O., Agyare, C. and Boakye, Y. D. (2015). Influence of methanol leaf extracts of *Hilleria latifolia* and *Laportea ovalifolia* on *in vitro* activity of selected antibiotics. 63<sup>rd</sup> International Congress and Annual Meeting of the Society for Plant and Natural Product Research (GA), Budapest Hungary. Absract number PW-177
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# DEDICATION

This work is dedicated to my parents, Very Rev. and Mrs. Oteng Dapaah. Indeed this is the fruit of their labour.

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# ABBREVIATIONS

AGEs	Advanced glycation end products
ALC	A merican type culture conection
AUC	Area under the curve
BHA	Butylated hydroxyanisole
BHT	Butylated hydroxytoluene
BrdU	5-bromo-2-deoxyuridine
CAT	Catalase
CFU	Colony forming unit
CoQ	Coenzyme Q
DMARDs	Disease modifying anti-rheumatic drugs
DMSO	Dimethylsulfoxide
DNA	Deoxyribonucleic acid
DPPH	2, 2-diphenyl-1-picrylhydrazyl
ECM	Extracellular matrix
EGF	Epidermal growth factor
FCS	Foetal calf serum
FGF	Fibroblast growth factor
FTC	Ferric thiocyanate
GPx	Glutathione peroxidase
НаСаТ	Human adult high calcium low temperature
HCL	Hydrochloric acid
HPLC	High performance liquid chromatography
IL-1	interleukin-1
IL-1β	Interleukin-1 <sup>β</sup>
IL-6	Interleukin-6
LDH	Lactate dehydrogenase
LDLs	Low-density lipoproteins
MBC	Minimum bactericidal concentration
MDA	Malondialdehyde
MDRIs	Multi-drug resistance inhibitors
MFC	Minimum fungicidal
MIC	Minimum inhibitory concentration

3-(4,5-dimethylthiazol-2-yl)-2,5- diphenyltetrazolium bromide Non-steroidal anti-inflammatory drugs
Phosphate buffer solution
Platelet derived growth factor
Reactive oxygen species
Superoxide dismutase
Thiobarbituric acid
Transforming growth factor
Tumor necrosis factor
Tumor necrosis factor-α
2,4,6-tri(2-pyridyl)-s-triazine

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### **CHAPTER 1**

## **1.0 Introduction**

Plants serve as sources of medicine and the use of plants to treat and prevent infectious diseases and other ailments is well accepted as part of culture and traditions of Africans and other parts of the world. This practice has been in existence since time immemorial and up to 80% of the population in developing countries depend on plants for medicinal purposes (Ayyanar and Ignacimuthu, 2011; Calixto, 2005) due to their availability, easy accessibility, low cost and minimum side effects (Akharaiyi, 2011; Tabuti *et al.*, 2003).

About 30% of medicines employed for therapeutic purposes have their source from natural products such as plants (Ramos *et al.*, 2008), making herbal medicine an integral part of primary health care delivery all over the globe (Akinyemi *et al.*, 2005). Also, an appreciable number of allopathic or orthodox medicines such as quinine, morphine and acetyl salicylic acid have their sources from natural products (Akharaiyi, 2011).

Modern medications produced from plant isolates have their development based on the traditional use of the plants (Cragg and Newman, 2013). For instance, the synthesis of metformin and other bisguanidine-type anti-diabetic drugs was based on the model of galegine, a pure compound isolated from *Galega officinalis*, which is used traditionally for the treatment of diabetes (Fabricant and Farnsworth, 2001). Also, the antimalarial drug, quinine was isolated from the bark of *Cinchona officinalis*, which had long been used for the treatment of fevers (Buss *et al.*, 2003). However, few plants have been scientifically assessed for their quality, efficacy and toxicity (Calixto, 2005).

Also microbial resistance to almost all existing antibiotics is on the increase (Nascimento *et al.*, 2000) and there are severe side effects associated with prolong use

of synthetic drugs for treatment of diseases. An Action must be taken to address these problems.

This study therefore sought to evaluate the antimicrobial, antioxidant, antiinflammatory and wound healing properties of *H. latifolia* and *L. ovalifolia*, and their cytotoxic effects, based on the claim as wound healing agents in folklore medicine.

## 1.1 Antimicrobial activity

## **1.1.1 Plants as source of antimicrobial agents**

The search for new antimicrobial agents is of great importance to scientist of this age due to the increase emergence of microbial resistance (Tenover, 2006). The antimicrobial agents search includes various sources such as the synthetic compounds as well as bioactive agents from natural products including aquatic microorganisms and medicinal plants (Agyare *et al.*, 2012).

Medicinal plants are a great source of antimicrobial agents and the idea of plants been used as potential drugs for treatment of infectious diseases was well accepted even before the discovery of microbes (Rios and Recio, 2005). Antimicrobial agents that originate from plants have numerous therapeutic potential and are effective in the treatment of infectious diseases, and are devoid of the side effects that often come with the use of synthetic antibiotics. Compounds isolated from plants have been developed into various medications which have aided in the management of bacterial, fungal and other microbial infections (Agyare *et al.*, 2006a).

Over the years a number of researches have been carried out to investigate the antimicrobial properties of medicinal plants and compounds isolated from these plants

all over the world. These include a research that tested 122 plant species for their antimicrobial properties in Argentina. It was reported that, *Punica granatum* and Allium sativum inhibited Aspergillus niger, Lithrea ternifolia, Cassia occidentalis, Heimia salicifolia and Punica granatum prevented the growth of Escherichia coli, whereas Tabebuia impetiginosa, Baccharis microdonta, Salvia gilliesii, Psidium guineense and Rosa borboniana inhibited Staphylococcus aureus (Anesini and Perez, 1993). Bakkiyaraj and Pandiyaraj (2011), reported on antimicrobial activity of Azadirachta indica, Portulaca oleracea, Euphorbia hirta, Gmelina asiatica and Santalum album againsts some food-borne microorganisms. Also, antimicrobial potential of *Centrosema pubescens* against a wide range of microorganisms was observed by Ekpo et al. (2011). Agyare et al. (2006b), reported the antimicrobial activity of some medicinal plants in Ghana, indicating that, methanol and petroleum ether extracts of Nauclea latifolia, Bridelia atroviridis and Zanthoxylum gilletii exhibited antimicrobial activity against all the test organisms, with *Pseudomonas* aeruginosa been the least susceptible to the extracts of the plants. A report by Fleischer et al. (2003) revealed that the leaves and seeds of Bixa orellana has broad spectrum antimicrobial activity.

## 1.1.2 Microbial resistance to antimicrobial agents

Naturally, bacteria have the genetic ability to develop resistance to therapeutic agents (Nascimento *et al.*, 2000). Microorganisms that were previously known to have been controlled by antibiotics are now resistant to these same antimicrobial agents (Levy and Marshall, 2004). The problem of resistance poses great threat to public health now than ever before, because of expanding multidrug resistance in a single organism which drastically restricts therapeutic options (Levy, 2005). This has compelled a lot of big

pharmaceutical companies to leave the antibiotic discovery field to produce more profitable medications for treating other chronic disease (Projan, 2003).

Resistance to antibiotics leads to ineffective treatment of infectious diseases and increases the risk of complications resulting in fatal outcomes (Harbarth *et al.*, 2007). Inappropriate diagnosis, drug counterfeit, use of antibiotics in food production and animal rearing, non-compliance and under dosing, and sometimes uncontrolled use of antimicrobial agents are factors that have contributed immensely to the overwhelming increase in the microbial resistant menace in current years (Adu *et al.*, 2014). Also the pathogenic bacteria themselves have ways and strategies of acquiring or developing resistance to antibiotics, and these include, active efflux of drugs, alteration of target sites and inactivation of antibiotics by producing enzymes that degrade them (Sibanda and Okoh, 2007).

## 1.1.3 Resistance modulation and antibiotic modifying activity of plants

Microbial resistance to antimicrobial agents has limited the use of known cheap but potent antibiotics (Ranjan *et al.*, 2012). This has necessitated the search for new potent antimicrobial agents to combat the threat posed by resistant microbes.

However, Coates *et al.* (2002) reported the emergence of cross resistance to newly identified antibiotics and other antimicrobial agents suggesting that the new antimicrobial agents may be rendered ineffective in the very near future. Even though rational use of these medicines can decrease the rate of resistance development, it cannot eradicate the emergence of resistant strains (Sibanda and Okoh, 2007). There is therefore the need to discover and develop new compounds that will target and block resistance mechanisms to help contain and treat infections from these resistant strains (Oluwatuyi *et al.*, 2004).

Research has shown that plants contain compounds with or without antimicrobial property that can cause resistant microorganisms to be susceptible to a previously ineffective antibiotic (Aiyegoro *et al.*, 2009). The curative effect of plant extracts in combination with previously ineffective antibiotics has been referred to as resistance modulating activity (Gibbons, 2004).

In recent years *in vitro* antimicrobial studies conducted by researchers have shown marked increase in antimicrobial activity of some antibiotics against some resistant strains when they are combined with plant extracts. These studies include; the effect of *Acanthospermum hispidum* extract on the activities of amoxicillin and ciprofloxacin (Adu *et al.*, 2011). Aiyegoro *et al.* (2009) also reported that *Helichrysum pedunculatum* leaf extracts in combination with penicillin G, amoxicillin, chloramphenicol, oxytetracycline, ampicilin, tetracycline, erythromycin and ciprofloxacin resulted in up to 60% increase in the activity of these selected antibiotics against test organisms. Also, compounds isolated from *Rosmarinus officinalis* potentiated the activity erythromycin against erythromycin effluxing strains of bacteria (Oluwatuyi *et al.*, 2004).

In other studies, modification of activity of antibiotics led to nullified or reduced activity. For instance, Adu *et al.* (2014), reported that the activity of amoxicillin and ciprofloxacin was inhibited against all test organisms in the presence of *Myristica fragrance*, whereas activity of tetracycline decreased by 32 fold.

# 1.1.4 Methods for evaluating microbial susceptibility to antimicrobial agents and Minimum inhibitory concentration determination

Antimicrobial activity of an extract or any antimicrobial agent can be investigated using diverse methods. Nevertheless the performance of all these methods employed in determining microbial susceptibility to an agent can be judged by the minimum

inhibitory concentration (MIC), hence making the MIC a 'golden standard' considered in susceptibility testing (Andrews, 2001).

The MIC can be defined as the least concentration of an antimicrobial agent that can inhibit the growth of microorganisms (Andrews, 2001; Mazzola *et al.*, 2009). It is mostly used to investigate the *in vitro* antimicrobial activity of drugs and also used as a measure to determine whether a microbial strain is resistant to an antimicrobial agent or not (Andrews, 2001).

The commonly used methods for microbial susceptibility testing and MIC determination are basically grouped into two (2), which are the diffusion and dilution methods. The diffusion method comprises of the agar well and the disc plate diffusion methods, whereas the dilution method involves the broth and the agar dilution methods.

## 1.1.4.1 Diffusion method

This method basically involves the diffusion of an antimicrobial agent of a known concentration from a well or a disc into an agar inoculated with a standard inoculum size of the test organism. The agent is said to exhibit an antimicrobial activity if it is able to prevent or inhibit the growth of the test microorganism, which is indicated as clear zones around the well or disc. The zone of growth inhibition gives an idea as to the efficacy of the agent on the test organism, ie the larger the zone of inhibition, the more potent the agent is against the test organism, and vice versa (Fabry *et al.*, 1998). From this method, the MIC is determined by plotting a graph of diameter of mean zones of inhibition against log of concentration. The antilog of the intercept on the log of concentration axis is taken as the MIC (Wiegand *et al.*, 2008).

The diffusion method can be said to be simple, reproducible and cost effective, but there are some drawbacks associated with it. The inhibition of growth by the agent in the well

or disc depend on the ability of the agent to diffuse through the solid medium (Scorzoni *et al.*, 2007a; Scorzoni *et al.*, 2007b), this makes it not a suitable method for the evaluation of antimicrobial property of non-polar agents and those that cannot diffuse readily through the agar. However, the potency of an antimicrobial agent is not solely dependent on its solubility and diffusion profile in the agar (Cos *et al.*, 2006).

## 1.1.4.2 Dilution method

In this method, different concentrations of the test antimicrobial agent are prepared (through dilution) in a suitable culture medium and then inoculated with test microorganism of specific inoculum size. In this method there is direct contact of the test antimicrobial agent and organism, so there is no need for diffusion of the test antimicrobial agent into the medium. The MIC is recorded after the period of incubation as the lowest concentration of the antimicrobial agent that inhibits growth of the test organism (Cos *et al.*, 2006).

Even though this method is said to be more reproducible and quantitative than the diffusion methods and is suitable for determining MIC, minimum bactericidal concentration (MBC) and minimum fungicidal concentration (MFC) of both polar and non-polar agents, the MIC that is recorded does not always denote the exact value.

The exact or actual MIC value lies between the least concentration that inhibited growth and the highest concentration that allowed growth (Cos *et al.*, 2006).

### 1.1.4.2.1 Agar dilution method

In the agar dilution method, a suitable solid culture medium is incorporated with antimicrobial agent at different dilutions and then inoculated with specific inoculum size of test organism. Agar is allowed to set and incubated in appropriate condition, after which growth is assessed in all the concentrations used and MIC determined. The

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MIC is taken to be the least concentration of the test antimicrobial agent that did not allow growth after the period of incubation (Cos *et al.*, 2006; Wiegand *et al.*, 2008).

## 1.1.4.2.2 Broth dilution method

The broth dilution method can be of two classes, depending on volume of liquid medium employed, that is, micro-dilution or macro-dilution. Micro-dilution technique uses a little amount of the broth, between 100 and 200  $\mu$ L and employs the use of micro-titre plate, while the quantity of broth employed in macro-dilution technique is between 1 and 2 mL (Wiegand *et al.*, 2008; Andrews, 2001). In this method, different concentrations of the test antimicrobial agent are prepared in the liquid medium and standard inoculum size of microorganism added to it. This is followed by incubating at the required temperature and time. Turbidity, indicating microbial growth in broth is assessed at the various concentrations and MIC is determined (Wiegand *et al.*, 2008; Lourens *et al.*, 2004). The MIC value is recorded as lowest concentration or the highest dilution of the test agent that inhibited the growth of test organism.

Microbial growth observation can be aided by addition of tetrazolium salts or resazurin dye after the period of incubation, which causes a colour change to indicate growth or no colour change to indicate no growth (Andrews, 2001; Umeh *et al.*, 2005). In the micro-dilution technique, the use of plate reader to measure absorbance can also be employed to determine the extent of microbial growth in each concentration and the MIC (Devienne and Raddi, 2002; Matsumoto *et al.*, 2002). Employing the plate reader, MIC values are determined as the least concentration of test agent which completely (100%) inhibited growth of microbes (Marchal *et al.*, 2013).

## **1.2 Free radicals and reactive oxygen species (ROS)**

Free radicals are molecules that can exist independently for a brief period and are made up of one or more highly reactive unpaired electrons, which reacts with other molecule by accepting or donating electron (Cui *et al.*, 2004; Valko *et al.*, 2007). ROS are collective term for all forms of activated oxygen or reactive oxygen, which include free radicals like ozone (O<sub>3</sub>) and molecular oxygen such as superoxide ( $O_2^-$ ) and hydroxyl (OH<sup>\*</sup>) (Cui *et al.*, 2004; Im Kim *et al.*, 2010).

ROS are continuously generated in the body of humans through the process of aerobic respiration and from other exogenous sources (Narayanaswamy and Balakrishnan, 2011). These ROS when produced are acted upon and detoxified by endogenous enzyme antioxidants. However due to their high chemical reactivity, when they are generated in excess or there is inadequate antioxidants defense, they can lead to oxidative damage of biomolecules such as proteins, lipids and DNA (Cui *et al.*, 2004; Farber, 1994).

Oxidative stress, that is referred to as the imbalance between the production of free radicals and opposing antioxidant defense mechanism, implicates a lot of diseases like diabetes mellitus, cancer, atherosclerosis, hypertension, ocular diseases, haematological diseases, inflammatory diseases, pulmonary diseases and neurological diseases (Cui *et al.*, 2004).

## 1.2.1 Antioxidant activity

Antioxidants may be referred to as intracellular or extracellular substances produced to counter the activity or effects of reactive oxygen species (ROS). They are used by organisms as part of their defense mechanism against the attack of free radicals. Some benefits of antioxidants include their use as stabilizers in polymeric product, petrochemicals, foodstuffs, pharmaceuticals and cosmetics (Pisoschi and Negulescu, 2011). Also they are used in relation to radical production and oxidative stress, cancer prophylaxis and therapy (Kalcher *et al.*, 2009).

Antioxidants may be grouped into three (3) based on their mechanism of action and chemical nature. These are; enzyme, preventive and scavenging or chain-breaking antioxidants (Cui *et al.*, 2004).

Enzyme antioxidants are endogenously produced to act against specific ROS and break them down to less harmful forms. Examples are superoxide dismutase (SOD) which converts superoxide radical to hydrogen peroxide, glutathione peroxidase (GPx) which detoxifies hydrogen peroxide by reducing it to water, and catalase (CAT) which also detoxifies hydrogen peroxide by decomposing it to water and oxygen (Cui *et al.*, 2004).

Preventive antioxidants are extracellular and they act by binding to and sequestering molecules which promote oxidation and transition metal ions that strongly enhance formation of free radical due to the presence of unpaired electrons. Examples include transferrin and lactoferrin which bind ferric ions, ceruloplasmin (which binds Cu, catalyzes the oxidation of ferrous ions to ferric due to its ferroxidase activity and increases the binding of iron to transferrin), haptoglobins which bind hemoglobin, hemopexin which binds heme, and albumin which binds copper and heme (Cui *et al.*, 2004).

The scavenging or chain-breaking antioxidants are lipid-soluble and water-soluble scavengers and they act by availing themselves for oxidation at the early stages of free radical chain reaction, thereby forming low energy products that cannot propagate the chain further. Major lipid-soluble scavengers include vitamin E (a-tocopherol),

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hcarotene, and coenzyme Q (CoQ) (Murthy, 2001), whereas water-soluble scavengers consist of ascorbic acid, uric acid, and bilirubin (Frei *et al.*, 1988).

## **1.2.2 Medicinal plants as source of antioxidants**

Phenolic compounds and flavonoids widely distributed in plants have been reported to exert numerous biological activities and have also been known as the main antioxidant compounds of fruits and vegetables (Wu and Ng, 2008).

Herbs, spices and other plant material rich in phenolic compound have been extracted and are widely use in the food industries since they are able to improve the quality and nutritional value of foods by retarding oxidative degradation of lipids (Patel *et al.*, 2010).

Recently, there is an increasing interest in the search for therapeutic potential of plants as antioxidants, due to adverse side effects and toxicity of existing synthetic antioxidants such as butylated hydroxyanisole (BHA) and butylated hydroxytoluene (BHT) (Qader *et al.*, 2011).

Medicinal plants containing antioxidant compounds (phenolics and flavonoids) which protects cells from the damaging effects of reactive oxygen species have already been exploited for commercial use as either antioxidant additives or as nutritional supplements, while other plant species are been investigated for their antioxidant properties. A lot of researchers have been involved in the search for new and novel antioxidants from plant source in recent years. These researches include; the antioxidant property of *Artocarpus Altilis* (Amponsah *et al.*, 2014). Mensah *et al.* (2004) reported on the antioxidant activity of *Commelina diffusa*. Whereas the antioxidant properties of *Kigelia africana* and *Strophanthus hispidus* have been reported by Agyare *et al.* (2013a).

## 1.2.3 In vitro assays used to determine antioxidant activity

The *in vitro* assay methods for assessing the antioxidant activity of a product can be grouped into two (2), namely direct and indirect methods. Indirect methods do not involve oxidisable substrate and are used to measure the ability of the presumed antioxidant molecule to reduce a stable artificial free radical or a transition metal either by hydrogen or electron transfer. Example of assays that fall under indirect methods include, 2, 2- diphenyl-1-picrylhydrazyl (DPPH) scavenging assay, total antioxidant capacity test and ferric-reducing antioxidant power assays(Laguerre *et al.*, 2007).

Direct methods involve an oxidisable substrate and are used to assess the potential of the antioxidant to inhibit the oxidative degradation of the substrate in a test system subjected to natural or accelerated oxidation conditions. The oxidisable substrate normally include lipids, plant proteins, fluorophores, chromophores, DNA, fluids such as low-density lipoproteins (LDLs) and biological membranes. Example of such methods is linoleic acid autoxidation method (Laguerre *et al.*, 2007).

## 1.2.3.1 DPPH free radical scavenging assay

The principle behind this method is based on electron transfer or hydrogen atom transfer reaction (Huang *et al.*, 2005). In this method, the free radical scavenging ability of an antioxidant compound is assessed in the presence of DPPH which is a stable free radical. The DPPH which has a violet colour is reduced to a yellow chromophore in the presence of the antioxidant. After 30 min of reaction, it is expected that the supposed antioxidant substance should mop up the DPPH free radicals in solution and excess free radicals measured using spectrophotometry by taking absorbance usually at 517 nm (Prior *et al.*, 2003; Badarinath *et al.*, 2010; Khalaf *et al.*, 2008).

The percentage DPPH free radical scavenging activity is calculated afterwards and used to determine the concentration that caused 50% reduction in the concentration of DPPH free radicals in solution, referred to as  $IC_{50}$  or  $EC_{50}$  (Badarinath *et al.*, 2010). This assay is very simple and rapid, but may have some few drawbacks. Compound such as carotenoids can appreciably absorb at 515 to 517nm, hence can influence spectrophotometry readings and affect the outcome of the experiment. Again, the antioxidant activity can be affected by the antioxidant's chemical structure (steric hindrance), small compounds can easily access and scavenge DPPH free radical, while bulky structures can be hindered or have difficulty in accessing free radicals, thereby appearing to have no or reduced antioxidant activity (Prior *et al.*, 2003).

## 1.2.3.2 Total antioxidant capacity assay

This assay involves the reduction of molybdenum by an antioxidant.  $Mo^{+6}$  is reduced to  $Mo^{+5}$  in the presence of an antioxidant compound. This reduction leads to the formation green phosphate-molybdate V ( $Mo^{+5}$ ) complex at acidic pH, which is spectrophotometrically quantified by measuring absorbance at 695 nm (Prieto *et al.*, 1999). The higher the absorbance of the green phosphate-molybdate V ( $Mo^{+5}$ ) complex formed, the greater the antioxidant activity of the compound and vice versa. This assay is quantitative, as it expresses the antioxidant activity as equivalence of ascorbic acid or  $\alpha$ -tocopherol.

## 1.2.3.3 Total Phenol (Folin-Ciocalteu) Assay

This assay involves the reduction of phosphomolybdate phosphotungstate salts of Folinciocalteau by phenolic compounds in a basic medium. The reduced form of the salt gives a blue colouration and is quantified spectrophotometrically at 760 nm (Slinkard and Singleton, 1977; Socha *et al.*, 2009). This assay is also quantitative and it expresses the phenol content as tannic acid equivalent.

## 1.2.3.4 Ferric-reducing antioxidant power assay

This assay is based on the principle of reducing ferric (Fe<sup>3+</sup>) to ferrous (Fe<sup>2+</sup>) by an antioxidant compound. The resultant Fe<sup>2+</sup> reacts with 2,4,6-tri(2-pyridyl)-s-triazine (TPTZ) to form a TPTZ-Fe<sup>2+</sup> complex which gives an intense blue colour. Absorbance is measured at 593 nm to quantify the amount of reduced iron which is related to the antioxidant activity of the compound (Benzie and Strain, 1996; Pisoschu and Negulescu, 2011).

## 1.2.3.5 Lipid peroxidation inhibition assay

This assay involves peroxidation of polyunsaturated fatty acid (linoleic acid) at a temperature of 40°C. Products formed at the primary and secondary stages of the degradation are used as a measure of the extent of lipid peroxidation, by employing ferric thiocyanate (FTC) and thiobarbituric acid (TBA) methods (Inatani *et al.*, 1983).

In the ferric thiocyanate (FTC) method, peroxides produced at the primary stage of degradation oxidize  $Fe^{2+}$  (blue) to  $Fe^{3+}$  (reddish brown), which is then quantified by forming a complex with SCN<sup>-</sup> ions (in ammonium thiocynate) and absorbance measured at 500 nm. In the presence of an agent with a high antioxidant activity, the amount of peroxides produced during peroxidation of linoleic acid is low, thereby low formation of  $Fe^{3+}$  ions and low absorbance values recorded.

In the thiobarbituric acid (TBA) method, malondialdehyde (MDA) formed by degradation of peroxides at the secondary stage is measured. Malondialdehyde (MDA) binds to thiobarbituric acid (TBA) at a low pH and high temperature (100°C) to form

TBA-MDA adduct, which gives a pink colouration that is correlated to the extent of lipid peroxidation by measuring absorbance at 532 nm. The higher the antioxidant activity, the less the intensity of the pink complex formed and hence the lower the absorbance values (Yin *et al.*, 2011; Pisoschu and Negulescu, 2011).

## **1.3 Inflammation**

Inflammation is a complex biological mechanism employed by vascular tissue for repair when injured or when exposed to harmful stimuli like pathogens or infections, irritants and physical injuries. It is a protective action that an organism uses to remove injurious stimuli, initiate healing process, as well as generate new tissues to replace damaged ones (Singh *et al.*, 2008; Schmid-Schönbein, 2006).

Classical signs associated with inflammation include, pain, redness, swelling, heat and loss of function (Shailasree *et al.*, 2012; Kumar *et al.*, 2010; Rock and Kono, 2008). An organism's ability to mount an inflammatory response is very necessary for its survival when injured or when infected by pathogens and other harmful stimuli.

Inflammatory response can occur in three distinct temporal phases; (a) an acute phase which involves vasodilation and increased capillary permeability; (b) sub-acute phase, where phagocytic cell and leukocytes migrate to the inflammation site; and finally (C) chronic proliferative phase characterized by tissue degeneration and fibrosis (Burke *et al.*, 2006).

The inflammatory response can only be established in the presence of inflammatory inducers and mediators. Inflammatory inducers are signals that initiate inflammation, while inflammatory mediators are substances with biological properties that cause or enhance the signs and symptoms of inflammation.

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Inducers of inflammation may be endogenous and exogenous. Endogenous inflammatory inducers can be signals from stressed, malfunctioning or damaged tissues and may also include crystals of monosodium urate and calcium pyrophosphate dihydrate, advanced glycation end products (AGEs) and oxidized lipoproteins (Rock and Kono, 2008). Exogenous inducers consist of microbial and non-microbial signals. Microbial inducers can be pathogen-associated molecule patterns or virulence factors, while non-microbial inducers include allergens, irritants, foreign bodies and toxic compounds (Rizki and Rizki, 1992; Dostert *et al.*, 2008).

Mediators of inflammation can be derived from plasma and blood cells or tissues. There are four major mediator-producing systems contained in plasma which interact in defined manners to generate phlogistic compounds. These include kinin, complement, coagulation and fibrinolytic systems, which are inactive proteases in their native form but are activated by proteolytic cleavage leading to the rise of more inflammatory mediators (Rang, 2007).

Cell or tissue mediators may be preformed and stored in granules or may be newly synthesized by cells. Preformed cell mediators may include histamine in mast cells and cationic proteins in neutrophils, while newly synthesized cell mediators may comprise of interleukin-1, leukotrienes, platelet-activating factor (Larsen and Henson, 1983).

Inflammation can be classified as acute or chronic, depending on the nature of the stimulus and the extent to which the stimulus was eliminated by the initial processes. Acute inflammation is an initial response of the body to harmful stimuli, which is rapid lasting for hours or few days. It can be induced by injury, infections and microbial toxins, tissue necrosis, hypoxia and immune reactions (Kumar *et al.*, 2010) and is characterized by increased release of plasma and leukocytes from the blood to the

injured area (Shailasree, *et al.*, 2012). This leads to the successful elimination of the harmful stimuli and complete healing of the endangered tissue. However, if the acute response is not able to eliminate the stimuli, it can result in the chronic state.

Chronic inflammation results from either the persistence of the injurious agent or some interference with the normal process of healing leading to prolonged duration of inflammation (weeks or months). It is characterized by the production of mononuclear cells including macrophages, lymphocytes and plasma cells as well as tissue destruction. Chronic inflammation may follow acute inflammation, or may begin insidiously as a low-grade smouldering response without any manifestations of an acute reaction, leading to tissue damage in most of the common and disabling human diseases like rheumatoid arthritis, atherosclerosis, tuberculosis and pulmonary fibrosis. This type of chronic inflammation may also be implicated in the progression of cancer and Alzheimer disease (Kumar *et al.*, 2010).

## **1.3.1 Management of inflammation**

Though the inflammatory response is a defensive mechanism on its own, it may lead to serious tissue damage which may call for an intervention to enhance the healing process (Obiri *et al.*, 2013). This calls for the use of principal orthodox antiinflammatory drugs such as non-steroidal anti-inflammatory drugs (NSAIDs), steroidal anti-inflammatory drugs (glucocorticoids) and disease modifying antirheumatic drugs (DMARDs). These drugs though effective, have been reported to be associated with severe side effects (Graham, 2006), which has necessitated the search for new and effective anti-inflammatory drugs with no or little adverse effects. The search is ongoing and can never be completed without investigating into natural products especially medicinal plants since they have been known as a great store of therapeutic agents (Cragg and Newman, 2013).

Medicinal plants have been promising when it comes to handling actions that are likely mediated through interference with the predominant pathophysiological processes of inflammation (Obiri *et al.*, 2013). Examples of such plants with antiinflammatory activity include fruits of *Xylopia aethiopica* (Obiri and Osafo, 2013), aerial parts of *Hilleria latifolia* (Abotsi *et al.*, 2012), *Ulva lactuca* (sea weed)

(Margret *et al.*, 2009), whole plant of *Sarcostemma secamone* (Kumari *et al* 2012), Stem bark and leaves of *Pterygota macrocarpa* and *Cola gigantea* (Agyare *et al.*, 2012), leaf and bark extracts of *Funtumia elastica* (Agyare *et al.*, 2013b).

### 1.3.2 Models used to evaluate anti-inflammatory activity of an agent

A number of animal models can be employed in both acute and chronic inflammation models to assess the anti-inflammatory activity of an extract or a compound.

### **1.3.2.1 Acute inflammation model**

Acute inflammation may be induced by injecting various parts of an animal's body with an inflammatory agent such as killed bacteria (e.g. *Escherichia coli*), carrageenan, formalin, polymorphonuclear leucocytes, arachidonic acid (in acetone), zymosan and monosodium urate crystals (Issekutz and Issekutz, 1989; Higgs, 1989). The animal responses to such inflammatory agents and this can be monitored by increase in foot volume (oedema in rat's paw), local rise in temperature, detection of plasma markers in skin, measurement of inflammatory mediators in plasma exudates, monocytes infiltration, leucocytes and lymphocytes accumulation, and platelet deposition using various techniques (Issekutz and Issekutz, 1989).

The most widely used and acceptable preliminary animal model of acute inflammation employed in the screening of new anti-inflammatory drugs is the carrageenan-induced acute foot oedema in laboratory animals (Winter *et al.*, 1962). Carrageenan which is a polysaccharide derived from the Irish sea moss is injected into the foot of laboratory animal to induce inflammation with a characteristic swelling paw (oedema) (Morris, 2003). The animal is treated with the test drug and the volume of the oedema is assessed to ascertain the anti-inflammatory activity of the drug.

Carrageenan (phlogistic agent) is widely accepted and used in the acute inflammation model because it is known to be antigenic and is devoid of apparent systemic effect (Di Rosa, 1972; Kaur *et al.*, 2004). It is also known to cause the release of more than one inflammatory mediator which is a useful tool in assessing anti-inflammatory effect of an agent. Again it has the ability to induce an intense and reproducible inflammatory action and it is sensitivity to inhibition by various anti-inflammatory drugs (Winter *et al.*, 1962; Kaur *et al.*, 2004).

## 1.3.2.2 Chronic inflammation model

A lot of models have been developed for the study of chronic inflammation. These include the polyarthritis induced by type II collagen in rats and mice (Trentham *et al.*, 1977), the arthritis induced by streptococcal cell walls in the rat (Cromartie *et al.*, 1977), chronic inflammation induced by implantation of cotton wool pellets subcutaneously (cotton pellet-induced granuloma) (Winter and Porter, 1957). Also, injection of turpentine oil (into pleural cavity or subcutaneous pouch) (Selye, 1953, Robert and Nezamis, 1957) and the most frequently studied model of polyarthritis induced in the rat with *Mycobacterium tuberculosis* (Freund's adjuvant arthritis) (Whicher *et al.*, 1989).

Models of arthritis is employed extensively in research because of its reproducible predictiveness regarding the activity of NSAIDs, even though it is not yet useful in
finding anti-arthritic agents with disease modifying properties, and because they closely resemble chronic inflammatory disease in man (Weichman, 1989).

# 1.4 Wounds

Wounds are physical, chemical or thermal break in continuity of the skin or its underlying tissues leading to disturbance in the normal skin anatomy and disruption of anatomical and functional integrity of living tissues (Singh *et al.*, 2006). Wounds normally serve as a passage or entry for infectious microbial agents into the inner tissues (Krasner *et al.*, 2007; Ofori-Kwakye *et al.*, 2011; Singh *et al.*, 2006).

Wounds may be acute or chronic depending on the healing process, and open or closed based on the cause and nature of the wound.

#### 1.4.1 Acute wounds

Acute wounds are wounds that go through the normal phases of healing thereby showing signs of healing in some few weeks. In acute wounds the processes involved in healing is always completed within the expected time frame. There is a great risk of infection within the first 72 hours of wound occurrence and for that matter proper care must be ensured within this period. Examples of acute wounds may include postoperative wounds and cuts (McGuckin *et al.*, 2003).

## 1.4.2 Chronic wounds

Chronic wounds are those that exhibit impaired healing and have failed to go through the normal phases of healing within four weeks, thereby by showing uncoordinated, postponed and incomplete healing. That is, chronic wounds take a long time to heal, do not heal or recur more often. Examples of chronic wounds are ulcers associated with diabetes mellitus, ischemia and pressure ulcers (Guo and DiPietro, 2010; McGuckin *et al.*, 2003; Krishnan, 2006).

#### 1.4.3 Open wounds

Open wounds lead to break or crack in the skin exposing the underlying tissues to the outside environment. External bleeding is mostly observed and the risk of microbial infection is very high. Examples of open wound are punctures, lacerations, incisions, abrasions and avulsions (Lazarus *et al.*, 1994; Nagori and Solanku, 2011).

# 1.4.3.1 Punctures

These are open wounds caused by small pointed objects such as tip of a knife, needle, a nail or a sharp tooth. Bleeding in such wounds is very minimal and are therefore at a higher risk of infection since bleeding leads to the removal of foreign materials that may enter into the depths of the wound.

#### 1.4.3.2 Lacerations

This type of open wounds is caused by some blunt trauma resulting in tissue damage or lost. Mostly they form lines on the surface of the skin but can extend into the lower tissues.

#### 1.4.3.3 Incisions

Incised wounds are caused by sharp-edged object such as a knife, razor or glass splinter and are characterized by no tissue loss and minimal tissue damage. Bleeding may be profuse hence immediate attention is needed.

# 1.4.3.4 Abrasions

Abrasions are caused when the surface of the skin gets rubbed against a rough or hard surface. Normally the topmost layer of the skin is scrapped off and the nerve endings are exposed, leading to shallow and painful injuries.

#### 1.4.3.5 Avulsions

These are type of injury whereby a part of the body is forcefully removed from its original place. An example of this type of injury is seen in an amputation where the body part is pulled off rather than cut off. Bleeding is very severe and due to its seriousness nature, immediate attention is needed.

# 1.4.4. Closed wounds

These are injuries that cause damage to internal tissues and organs without any break in the outermost layer of the skin. In this type of injury, blood escapes the circulatory system but remains inside the body. The risk of microbial contamination is very minimal in closed wounds since they are not exposed to the outside environment. Examples of such wounds include bruises, haematomas and crush injury (Lazarus *et al.*, 1994; Nagori and Solanku, 2011)

## 1.4.4.1 Haematomas

This type of closed wound is also known as blood tumor and is caused by damage to a blood vessel which results in blood been collected under the skin.

#### 1.4.4.2 Contusion or bruises

These are injuries that cause damage to tissues under the skin and are caused by blunt force trauma.

# 1.4.4.3 Crush injury

Crush injuries are caused when a great or extreme amount of force is applied on the skin for a long period of time.

#### 1.4.5 Process of wound healing

Wound healing is a complex process instigated immediately after injury to repair damaged tissues or organs. The process consist four (4) distinct but interconnected and overlapping phases which occur in proper sequence and time frame to enhance successful healing (Guo and DiPietro, 2010; Nagori and Solanku, 2011). These integrated phases are haemostasis (coagulation), inflammation, proliferation and tissue remodelling (Gosain and DiPietro, 2004).

## 1.4.5.1 Haemostasis phase

This phase starts right after injury or wounding resulting in vascular constriction and fibrin clot formation. Platelets aggregation at the site of injury leads to the production of thrombin and certain factors which cause the intrinsic clotting cascade to convert fibrinogen to fibrin, which forms a mesh (clot) to control active bleeding and helps to achieve haemostasis. The clot and surrounding wound tissue also lead to release of pro-inflammatory cytokines and growth factors like transforming growth factor (TGF)- $\beta$ , fibroblast growth factor (FGF), epidermal growth factor (EGF) and plateletderived growth factor (PDGF) to initiate the next phase (Guo and DiPietro, 2010).

#### 1.4.5.2 Inflammatory phase

This phase starts with the migration of pro-inflammatory cells to the wound site within 6 to 8 hours after bleeding has been controlled. The inflammatory phase is characterised by vasodilation of blood vessels and infiltration of neutrophils, which act to clear any invading microorganisms and cellular debris at the wound site. Macrophages are subsequently released to continue the degradation of microbes and other tissue debris at the wound area. Also macrophages cause the release of growth factors to enhance

healing in the proliferative phase, and lead to the multiplication of endothelial cells and smooth muscle cells. They again secrete other factors like interleukin-1 (IL-1) and tumor necrosis factor (TNF) which facilitate the healing process (Habif, 1996). Macrophages clear apoptotic cells (including neutrophils), leading to subsequent stimulation of keratinocytes, fibroblasts and angiogenesis to promote tissue regeneration (Meszaros *et al.*, 2000; Mosser and Edwards, 2008).

## 1.4.5.3 Proliferative phase

This phase normally follows and overlaps with the inflammatory phase lasting for 2 days to 3 weeks after the inflammatory phase. At this phase fibroblasts and endothelial cells are the most prominent cells present and enhance capillary growth, collagen formation, and the formation of granulation tissue at the injury site. The proliferation phase consists of three main steps, which are granulation, contraction and epithelialization. In the granulation stage, fibroblast form collagen bed and produce new capillaries. Within the wound bed, fibroblasts produce collagen as well as glycosaminoglycans and proteoglycans which are major components of the extracellular matrix (ECM) and are essential for wound repair. At the contraction stage, pulling together of wound edges to reduce the defect is observed. During the epithelialization stage, new and fresh epithelial tissues form over the wound site, with cell migrating from the periphery of the wound and adnexal structures (Guo and DiPietro, 2010; Nagori and Solanku, 2011). Clinically, proliferation is characterized by pebbled red tissues or collagen formation in the wound base and leads to the replacement of dermal tissues and in few instances sub-dermal tissues (in deep wounds), along with wound contraction (Stadelmalmann et al., 1998; Kane, 2007).

#### 1.4.5.4 Tissue remodelling phase

The remodeling phase can last for years and is generally characterized with extracellular matrix (ECM) remodelling to approach the normal or original tissue. In this phase, regression of many of the newly formed capillaries takes place, so as to cause the vascular density of the wound to return to normal, as well as physical contraction of wounds believed to be a function of contractile fibroblasts (myofibroblasts). Also intermolecular cross-linkage of collagen leads to increased tensile strength of tissue, with scar tissue becoming about 80% stronger as the original tissue (Guo and DiPietro, 2010; Nagori and Solanku, 2011).

# 1.4.6 Medicinal plants as source of wound healing agents

Orthodox medications have been employed to shorten the duration of the healing process and minimize complications associated with natural wound healing (such as wound infection due to bacterial contamination). However, the cost of these orthodox medications are outrageous and wound contaminating bacteria are becoming more resistant to their effect (Udegbunam *et al.*, 2014). The search for newer, effective and affordable alternatives has become the focus of a lot of scientist now.

Medicinal plants have been widely used and accepted as more efficacious in traditional medicine for the treatment of wounds among past and present generations (Udegbunam *et al.*, 2014). The following are some medicinal that have been reported with wound healing proprties. *Pupalia lappacea* (Apenteng *et al.*, 2014; Udegbunam *et al.*, 2014), *Spathodea campanulata* (Ofori-Kwakye *et al.*, 2011), *Kigelia pinnata* (Sharma *et al.*, 2010), *Carica papaya* (Mahmood *et al.*, 2005), *Myrianthus arboreus* and *Alchornea cordifolia* (Agyare *et al.*, 2014), *Rubus sanctus* (Süntar *et al.*, 2009), *Aloe vera* (Yadav *et al.*, 2012; Oryan *et al.*, 2010), *Justicia flava, Lannea welwitschii* 

(Agyare *et al.*, 2013c), *Strophanthus hispidus* (Agyare *et al.*, 2013a) and *Clerodendron slendens* (Gbedema *et al.*, 2010).

#### 1.4.7 Models for assessing wound healing activity of an agent

A lot of models have been developed to evaluate the influence of an agent on the healing processes and these models can be classified as *in vitro* and *in vivo*. For any of the models to be employed in a study, a number of factors must be considered, especially the parameters to be measured.

#### 1.4.7.1 In vitro wound healing models

The *in vitro* models involve minimal ethical consideration, are very rapid and simple, and are cost effective as compared to the *in vivo* models. The *in vitro* models available for assessing the wound healing activity of an agent include single cell systems, multicellular systems and organ cultures (intact skin). These models can be employed in a study to mimic cell migration and proliferation, cell-cell and Cell-matrix interaction, protein synthesis, wound contraction and epithelialisation during wound healing (Gottrup *et al.*, 2000).

### 1.4.7.2 In vivo wound healing models

*In vivo* models employed in the study of the wound healing activity of an agent can grouped into artificial and tissue models. Artificial models consist of subcutaneous chamber or sponges and subcutaneous tubes, while tissue models include excisional wounds, incisional wounds, superficial wounds, dead space and burn wounds (Dorsett-Martin and Wysocki, 2008; Gal *et al.*, 2008).

Moreover models such as rabbit ear chamber, the hamster check pouch, the rabbit corneal pocket and the chick chorioallantoic membrane can be employed to assess parameters such as epithiliasation, neovascularization and dermal reconstitution (Gottrup *et al.*, 2000).

# 1.5 In vitro cell toxicity studies

The possible side effects and toxicity of medicinal plant and other compounds used for the treatment of diseases is an important part of new drug discovery and these cannot be overlooked. This has led to the development of various cytotoxicity assays including 3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT) assay to measure cell viability, 5-bromo-2-deoxyuridine (BrdU) assay to measure cell proliferation and Lactate dehydrogenase (LDH) release assay to assess the integrity of the cell membrane (Döll-Boscardin *et al.*, 2012).

# 1.6 Plants under study

#### 1.6.1 Hilleria latifolia (Lam.) H. Walt.

# Description

*Hilleria latifolia* (Lam.) H. Walt belongs to the family Phytolaccaceae and is locally known as '*Avegboma'*, '*Boe'* or '*Kukluigbe'* by the Ewes and '*Anafranaku'* by the Asantes. It is a perennial herb of 30 to 120 cm high. The leaves are ovate-elliptic, 15 cm long and 6 cm broad, obtuse or sub-acute at the base, with numerous short hairlike structures on lower surface. Petiole is 3 to 7 cm long whilst flowers are pink or white in slender racemes up to 13 cm long. The fruits are reticulate, glabrous and ellipsoid-globose

(Mshana, 2000; Dokosi, 1998).



Figure 1.1: Aerial parts of Hilleria latifolia Geographical

# distribution

*H. latifolia* is common on cultivated grounds and along forest paths in Ghana. It also occurs in other parts of tropical Africa as well as South America. It can also be found along the West African cost, from Sierra Leone to Cameroun (Dokosi, 1998).

# **Medicinal uses**

The leaves are used for the management of rheumatism, boils and wounds in Ghana (Agyare *et al.*, 2009). The leaves, added to those of *Piper guineense*, are applied to the body for the treatment of general oedema. In Congo, the leaves are employed to treat some skin diseases (Mshana, 2000; Dokosi, 1998).

Leaf decoction is taken or administered by enema to treat ascites and food poisoning as it causes violent purging. Also it is taken to treat coughing of blood (Schmelzer and Gurib-Fakim, 2008). A paste made from grounded flowers is taken together with fresh orange juice in the treatment of asthma (Mshana, 2000).

# Other uses

*H. latifolia* is used as an indicator of good soil and climate condition for the cultivation of cocoa and coffee. The dried seeds are used as a relish to tortoise meat (Dokosi, 1998).

WJSANE

The blackened stem is used by the Maasai people of Kenya in drawing eyebrows (Schmelzer and Gurib-Fakim, 2008).

#### **Phytochemical constituents**

Preliminary phytochemical screening of the ethanolic extract of the aerial parts of *H*. *latifolia* showed the presence of saponins, tannins, glycosides, steroids, terpenoids as well as small amounts of flavonoids and alkaloids (Woode *et al.*, 2011). In another report, methanolic fraction of *H. latifolia* showed the presence of glycosides, coumarins and reducing sugars, as well as small amount of triterpens and sterols.

However, saponins, tannins, flavonoids and alkaloids were absent (Assob et al., 2011).

# **Biological activity**

The leaf extract of *H. latifolia* has been demonstrated to have anxiolytic and antidepressant-like effects, antimicrobial, antioxidant and anti-inflammatory properties, (Abotsi *et al.*, 2012; Assob *et al.*, 2011; Woode *et al.*, 2011), antinociceptive and some neurobehavioral properties (Woode and Abotsi, 2011).

#### 1.6.2 Laportea ovalifolia (Schumach.) Chew.

#### Description

Laportea ovalifolia (Schumach.) Chew belongs to the family Urticaceae. It is known by the Asantes as 'akyekyenwonsa', 'abrewa nom taa' or 'Kumasi otuo'. It is a herbaceous weed more often creeping than erect, densely covered with stinging hairs. The perennial stems are cylindrical, greenish to sometimes reddish or brownish in colour often prostrate with erect shoots (Chew, 1969). *L. ovalifolia* is of two varieties that is *L. ovalifolia* (male and female). Since they are related species, they have the same characters but different in structure (Yang *et al.*, 1996). *L. ovalifolia* (male) have big leaves and (female) possess small leaves (Essiett, *et al.*, 2011).



Figure 1.2: Leaves of Laportea ovalifoila

# **Geographical distribution**

*L. ovalifolia* is a tropical plant commonly found in swampy areas in Cameroon and other parts of the world in both dry and rainy seasons (Letouzey, 1968).

# **Medicinal uses**

The leaves of *L. ovalifolia* are used as haemostatic on cuts and wounds (Agyare *et al.*, 2009), whereas the fruits are used as a poison antidote (Bouch, 2004). The root boiled in water is taken to prevent excessive menstrual bleeding (Sofowora, 1996).

# Other uses

The people of Ibibio tribe in Nigeria use the leaves and tender shoots of the plant as pot herb or vegetables in soups (Etukudo, 2003)

# Phytochemical constituents of the plant

Phytochemical analysis of the leaves extract of *L. ovalifolia* revealed the presence of saponins, tannins, flavonoids, phlobatanins and cardiac glycosides. Anthraquinone was absent (Essiett, *et al.*, 2011).

## **Biological activity**

*L. ovalifolia* has been reported to possess antimicrobial, antihyperglycemic activity and it is also effective in reducing oxidative stress in diabetes (Okwulehie and Akanwa, 2013; Iffen and Usoro, 2010). It also has antidiabetic and hypolipidemic effects in alloxan induced diabetic rats (Momo *et al.*, 2006).

# 1.7 Main objective

The aim of this study was to determine the antimicrobial and resistance modifying activities, anti-inflammatory and *in vivo* wound healing activities of leaf and root methanol extracts of *H. latifolia* and leaf methanol extract of *L. ovalifolia*.

# 1.7.1 Specific objectives

- To prepare leaf and root methanol extracts from *H. latifolia* and leaf extract of *L. ovalifolia*.
- To determine the phytochemical constituents and HPLC profile of the extracts.
- To determine the antimicrobial properties and minimum inhibitory concentrations of the extracts.
- To determine the ability of the extracts to modulate the resistance of some organisms to selected antibiotics.
- > To determine the antioxidant properties of the extracts.
- > To determine the anti-inflammatory properties of the extracts.
- > To determine the wound healing potential of the extracts.
- > To evaluate the cytotoxicity effect of the extracts

# **CHAPTER TWO**

# 2.0 Materials and methods

# **2.1 Materials**

# Table 2.1.1 List of instruments and equipment

Instruments/Equipment/Materials	Manufacturer/source/city		
Lab mill machine	Christy and Norris, England		
Ultra-Turrax (T 25)	Janke and Kunkel, Labortenik, Germany		
laboratory sieve	Retsch, Haan, Germany		
Filter paper	Whatmann, UK		
Rotary evaporator (R-210)	Buch, Germany		
Hot air oven	Gallenkamp, UK		
Refrigerator	Sharp Corporation, UK		
Beakers	Fisher Scientific, Schwerte, Germany		
Conical flask	Fisher Scientific, Schwerte, Germany		
Test tubes	Fisher Scientific, Schwerte, Germany		
Electronic weighing balance	Ohaus Corporation, PB, NJ, USA		
Thermostatically controlled water bath	New Brunswick, Edison, N J, USA		
HPLC machine	Agilent, USA		
HPLC column	Phenomenex Jupitoer C18 300R 250 x 4.6 mm		
	5 microns		
Syringe filters (0.45 µm pore size)	Sartorius Stedim Biotech, Goettingen, Germany		
Frost free freezer	Mistral, UK		
Micropipettes	Fisher Scientific, Schwerte, Germany		
Pipette tips (10, 100, 1000 μL)	Sarstedt, Nümbrecht, Germany		
96 well microtitre plates	Sarstedt, Nümbrecht, Germany		
Laminar air flow cabinet Model T 22472	Skan AG, Allschwill, Switzerland		
Portable autoclave	Basildon Ltd.UK		
Incubator	Thermo Scientific, Asheville, NC, USA		
Microtitre Plate Reader (MTP-Reader)	Bio Tek Instruments GmbH, Germany		
Electronic calipers (Z22855,	Milomex Ltd, Bedfordshire, UK).		
Surgical scissors	J J Int. Instruments, Kerala, India		
Toothed forceps	J J Int. Instruments, Kerala, India		
Axioscop Zeiss Microscope	Carl Zeiss Microscopy, Thornwood, USA		
Cell Proliferation ELISA, (5-bromo-2-deoxyridine,	Roche Diagnostic GmbH, Mannheim, Germany		
BrdU) Test Kit			
Cytotoxicity Detection Kit (LDH), Colorimetric	Roche Diagnostics GmbH, Mannheim, Germany		

Culture medium	Manufacturer/Company/Place
Mannitol salt agar	Oxoid Ltd, Basingstoke, UK
Cetrimide agar	Oxoid Ltd, Basingstoke, UK
MacConkey agar	Oxoid Ltd, Basingstoke, UK
Sabouraud agar	Oxoid Ltd, Basingstoke, UK
Sabouraud broth	Oxoid Ltd, Basingstoke, UK
Nutrient agar	Oxoid Ltd, Basingstoke, UK
Nutrient broth	Oxoid Ltd, Basingstoke, UK
Bismuth sulphite agar	Oxoid Ltd, Basingstoke, UK
Koser's citrate	Oxoid Ltd, Basingstoke, UK
Methyl Red Voges Proskauer (MRVP) broth	Oxoid Ltd, Basingstoke, UK

# Table 2.1.2 List of culture media

Tryptone broth	Oxoid Ltd, Basingstoke, UK		
Peptone broth	Oxoid Ltd, Basingstoke, UK		
HaCaT-keratinocyte medium	PAA Laboratories Pasching, Austria		
Table 2.1.3 List of chemicals/ reagents and refer	rence drugs		
Chemicals/Reagents/Reference drugs	Manufacturer/source/city		
Methanol	Merck BDH, Poole, UK		
Ferric chloride (99.9%)	Sigma Aldrich, London, UK		
Lead acetate	Sigma Aldrich, London, UK		
Ammonia (99.9%)	Sigma Aldrich, London, UK		
Hydrochloric acid (36.5%)	Sigma Aldrich, Michigan, USA		
Sulphuric acid (98%)	Sigma Aldrich, London, UK		
Sodium hydroxide (97%)	Sigma Aldrich, London, UK		
Fehling's solution	GPR, BDH, Poole, UK		
Ethanol	Merck BDH, Poole, UK		
Chloroform (99.9%)	Merck BDH, Poole, UK		
Dragendorff's reagent (99.8%)	Sigma Aldrich, London, UK		
Acetic anhydride (99.9%)	Merck BDH, Poole, UK		
Glycerol (88%)	Merck BDH, Poole, UK		
Ciprofloxacin (98% HPLC grade)	Sigma Aldrich, Michigan, USA		
Ketoconazole (98% HPLC grade)	Sigma Aldrich, Michigan, USA		
Amoxicillin	Sigma Aldrich, Michigan, USA		
Erythromycin	Sigma Aldrich, Michigan, USA		
Tetracycline	Sigma Aldrich, Michigan, USA		
Ampicillin	Sigma Aldrich, Michigan, USA		
3-(4,5-dimethylthiazol-2-yl)-2,5 diphenyltetrazoliun	n Sigma Aldrich, Taufkirchen, Germany		
bromide (MTT) (97%)			
1, 1-diphenyl-2-picryl-hydrazyl (DPPH) (98%)	Sigma Aldrich, Taufkirchen, Germany		
α-tocopherol (95.5%)	Sigma Aldrich, Taufkirchen, Germany		
Tannic acid (98%)	Merck BDH, Poole, UK		
Folin Ciocalteau (2 M)	Sigma Aldrich, St. Louis, Missouri, USA		
Sodium carbonate (99.9%)	Sigma Aldrich, St.Louis, Missouri, USA		
Ammonium molybdate (99.9%)	Sigma Aldrich, St. Louis, Missouri, USA		
Disodium phosphate (99.9%)	Sigma Aldrich, St. Louis, Missouri, USA		
Lambda Carrageenan	Sigma Aldrich, St. Louis, Missouri, USA		
Aspirin (99% w/w HPLC)	Sigma Aldrich, St. Louis, Missouri, USA		
Ketamine hydrochloride	Pfizer, New York, USA		
Normal saline (0.9%)	Intravenous infusions, Koforidua, Ghana		
Silver sulphadiazine cream (1% w/w)	Ayrton drugs, Accra, Ghana		
Buffered Formalin (10%)	Sigma Aldrich, Michigan, USA		
Xylene (99%)	Merck, BDH Poole, UK		
Paraffin	Paraplast Plus, Sherwood, St. Louis, USA		
Haematoxylin (90%)	Sigma Aldrich, St. Louis, Missouri, USA		
Eosin (99%)	Sigma Aldrich, St. Louis, Missouri, USA		
Phosphate buffer	Sigma Aldrich, St. Louis, Missouri, USA		
Dimethyl sulfoxide (DMSO)	Applied Chem GmbH, Darmstadt,		
	Germany		
Foetal calf serum (FCS)	Invitrogen, Karlsruhe, Germany		
Triton-X 100	Sigma Aldrich, St. Louis, Missouri, USA		

#### 2.2 Methods

#### **2.2.1 Plant collection and extraction**

Plant parts of *H. latifolia* (leave and roots) and *L. ovalifolia* (leaves) were collected from Aburi in the Eastern region of Ghana on February, 2014. The plants were authenticated by Dr. Alex Asase of the Department of Botany, University of Ghana, and voucher specimen AA 63 and AA 71, respectively deposited in the Ghana Herbarium, Department of Botany, University of Ghana, Legon, Accra. The plant parts collected were washed thoroughly under running tap-water and dried under shade at a temperature of 25 to 28°C for two weeks, after which they were pulverized into coarse powder using the laboratory milling machine.

Eight hundred grams (800 g) each of the powdered plant materials were soaked or suspended in 2.5 L of the 70% v/v methanol. They were extracted with ultra-turrax under ice-cooling at a speed of 24000 rpm for 3 to 5 min, and then filtered using a laboratory sieve of mesh number 200 with aperture of 75 µm and Whatmann filter paper No.1. The filtrates were concentrated with the rotary evaporator at 40<sub>o</sub>C under reduced pressure and allowed to dry in the hot air oven at 40<sub>o</sub>C and then stored in air tight containers at 4-8<sub>o</sub>C in a refrigerator. The weight of the extracts were recorded and labeled as *Hilleria latifolia* leaf methanol extract (HLMR) and *Laportea ovalifolia* leaf methanol extract (LOML).

# 2.2.2 Preliminary phytochemical screening

HLML, HLMR and LOML and the powdered plant materials were subjected to qualitative phytochemical screening to identify the various secondary metabolites present, using the methods described by Sofowora, (1993), Trease and Evans (2002), Ayoola *et al.* (2008), Vaghasiya *et al.* (2011) and Usman *et al.* (2014).

#### 2.2.2.1 Test for Tannins

Twenty-five milliliters (25 mL) of water was added to 0.5 g of each of the powdered plant materials, HLML, HLMR and LOML and then boiled for 5 minutes. The solution was allowed to cool and filtered. Ten (10 mL) of water was added to 1 mL of the filtrate, followed by the addition of 5 drops of 1% lead acetate solution. The appearance of buff precipitate indicated the presence of tannins. The method was repeated using 1% w/v ferric chloride (FeCl<sub>3</sub>) solution and observed for the appearance of dark green or blueblack colouration indicating the presence of condensed or hydrolysable tannins respectively.

# 2.2.2.2 Test for flavonoids

An amount of 0.5g each of the powdered plant materials, HLML, HLMR and LOML was separately dissolved in distilled water and filtered. A strip of filter paper was dipped in the liquid extracts and allowed to dry and then exposed to ammonia solution. The formation of an intense yellow colour which turned colourless after 5 min exposure to hydrochloric acid indicates the presence of flavonoids.

# 2.2.2.3 Test for glycosides (general test)

An amount of 200 mg each of the dried powdered plant materials, HLML, HLMR and LOML was warmed with 5mL of dilute H<sub>2</sub>SO<sub>4</sub> for 2 minutes. Each solution was cooled and filtered. The filtrates were made alkaline by adding 2 to 5 drops of 20% NaOH, (red litmus paper turned blue), after which 1 mL of Fehling's solution A and B was added and heated on water bath for 2 minutes. The mixture was observed for the formation of a brick red precipitate which indicates the presence glycosides.

#### 2.2.2.4 Test for saponin glycosides

Fifty milligrams each of the powdered plant materials, HLML, HLMR and LOML was shaken with 5 mL distilled water and filtered into a test tube. The filtrate was shaken vigorously to produce a froth which does not break readily on standing. The persistence of the froth for about 10 minutes after addition of 0.5 mL 2 M HCL is an indication for the presence of saponins.

## 2.2.2.5 Test for alkaloids

Fifty milligrams each of the powdered plant materials, HLML, HLMR and LOML was extracted with ammoniacal alcohol (95% ethanol: ammonia 9:1), filtered and evaporated to dryness. The residue was extracted with 1% H<sub>2</sub>SO<sub>4</sub>, filtered and rendered distinctly alkaline with dilute NH<sub>3</sub> solution. Afterwards it was shaken with CHCl<sub>3</sub> and the chloroformic extract layer was separated and evaporated to get rid of the CHCl<sub>3</sub>. The residue was then dissolved in 1% H<sub>2</sub>SO<sub>4</sub> and 1 to 2 drops of

Dragendorff's reagent was added. An orange red precipitate produced is an indication of the presence of alkaloids.

# 2.2.2.6 Test for sterols

Fifty milligrams each of the powdered plant materials, HLML, HLMR and LOML was extracted with chloroform and filtered into tests tubes. Two (2) mL acetic anhydride was added to 2 mL of the chloroformic extract and then concentrated  $H_2SO_4$  was carefully added at the side of the test tube. An appearance of a bluishgreen colour at the interface is an indication of the presence of sterols.

#### **2.2.2.7 Test Terpenoids**

Fifty milligrams each of the powdered plant materials, HLML, HLMR and LOML was extracted with chloroform and filtered, followed by careful addition of concentrated H<sub>2</sub>SO<sub>4</sub>. A reddish-brown colouration at interface indicates the presence of terpenoids.

# 2.2.3 High performance liquid chromatography (HPLC) Profile of extracts

The HPLC analysis was performed to know the chemical profiling of the crude extracts, which will serve as a guide in identifying the plants. The profiles of the extracts were determined using an HPLC machine with a UV-detector set at a wavelength of 254 nm. The extract was first dissolved in methanol and filtered through a syringe membrane filter (pore size 0.45  $\mu$ m) into a vial and then inserted into HPLC. The running conditions included injection volume of 10  $\mu$ L, mobile phase of methanol:water (20:80 v/v), flow rate of 1 mL/min and pressure of 15 MPa.

All the chromatographic data were determined using Chrom Quest software.

# 2.2.4 Test microorganisms

The microorganisms that were used in all the antimicrobial studies were obtained from the microbiology laboratory of the Department of Pharmaceutics, Faculty of Pharmacy and Pharmaceutical Sciences, Kwame Nkrumah University of Science and Technology (KNUST), Kumasi, Ghana. They included, *Pseudomonas aeruginosa* ATCC 4853, *Escherichia coli* ATCC 25922, *Staphylococcus aureus* ATCC 25923, *Bacillus subtilis* NCTC 10073, *Streptococcus pyogenes* (clinical strain), *Candida albicans* (clinical strain) and *Klebsiella pneumonia* (clinical strain). The organisms were cultivated on their respective selective media and other biochemical tests were performed on them to confirm their authenticity. They were stored in 30% glycerol broth at -  $4^{0}$ C in a frost free freezer until needed, whereby  $100\mu$ L of the stock suspension was transferred into broth and incubated at  $37^{0}$ C for 24 h (sub-cultured) before use.

# 2.2.5 Determination of antimicrobial activity

# 2.2.5.1 Determination of MIC of extracts

The susceptibility and minimum inhibitory concentration (MIC) of HLML, HLMR and LOML were determined against the test microorganisms using the broth dilution method as described by Eloff (1998) and Agyare *et al.* (2012). Each well of the micro-titre plate (96 wells) was filled with 100  $\mu$ L of double strength nutrient broth,

20  $\mu$ L of 10<sup>6</sup> cfu/mL of the test organisms and 80  $\mu$ L of different concentrations of HLML, HLMR and LOML and reference drugs prepared with sterile distilled water to obtain a gradient concentration ranging from 1.56 to 100 mg/mL and 1.0 to 128  $\mu$ g/mL, respectively. Control wells were filled with broth only and broth and test organisms only. After 24 hours of incubation, 20  $\mu$ L of 1.25 mg/mL of 3-(4, 5dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT) was added to each well and observed for a purple colouration after incubating at 37°C for 30 min which indicated microbial growth. The least concentration of HLML, HLMR, LOML and reference drugs that did not show any colour change in the wells was recorded as the MIC.

#### 2.2.5.2 Antibiotic modifying activity

This assay was performed to determine the effect of a sub-inhibitory concentration of the extracts on the activity of some selected antibiotics. The broth dilution assay with few modifications as described by Adu *et al.* (2014) was employed. The antibiotics used included amoxicillin, erythromycin, ciprofloxacin, tetracycline and ampicillin.

The MICs of the antibiotics were first determined using concentrations ranging from 1 to 1024 µg/mL. Each of the 96 wells of the micro-titre plate was filled with 100 µL of double strength nutrient broth, appropriate volume of the antibiotics and 20 µL of  $10^6$  cfu/mL of the test organisms. The plate was incubated for 24 h at 37°C, after which 20 µL MTT was added to the wells and MIC determined as the lowest concentration at which no growth was observed, (that is no colour change from yellow to purple).

The MICs of the antibiotics were re-determined in the presence of sub-inhibitory concentration (5 mg/mL) of the extracts (HLML and LOML).

# 2.2.6 Determination of antioxidant activity

## 2.2.6.1 DPPH free radical scavenging assay

This method was employed to assess the ability of HLML, HLMR and LOML to scavenge the DPPH free radical. Different concentrations (15.6 to 1000 µg/mL) of HLML, HLMR and LOML and standard antioxidant ( $\alpha$ -tocopherol) were prepared in methanol. DPPH solution (20 µg/mL) was prepared with methanol in the dark. Three milliliters (3 mL) of DPPH solution was added to 1 mL each of HLML, HLMR, LOML and  $\alpha$ -tocopherol and incubated in dark for 30 mins. Control was prepared by adding 3 mL of DPPH solution to 1 mL methanol and treated in the same way as the test samples. Absorbance of excess DPPH was measured at a wavelength of 517 nm (Susanti *et al.*, 2007; Braca *et al.*, 2001). The experiment was performed in triplicates.

The percentage scavenging was calculated using the equation below:

% scavenging = [(Absorbance control - Absorbance test)/ Absorbance control]  $\times$  100

#### 2.2.6.2 Total phenolic content

The total phenolic content of HLML, HLMR and LOML were determined by the method described by Škerget *et al.* (2005), using Folin-Ciocalteu reagent with few modifications. HLML, HLMR, LOML and standard drug (tannic acid) with concentration range of 0.5 to 10 mg/mL and 0.0156 to 1 mg/mL, respectively, were prepared. A volume of 0.1 mL Folin-Ceucalteu reagent was added to 0.5 mL each of HLML, HLMR, LOML and tannic acid solution, followed by the addition of 2.5 mL sodium carbonate (2%) and incubated at room temperature for 20 min. Absorbance was taken at 760 nm. The experiment was performed in triplicates. A blank was prepared by adding all the reagents with the exception of extract/standard drug and treated likewise. The total phenolics were expressed as milligrams per gams of tannic acid equivalents (TAEs).

**2.2.6.3 Total antioxidant capacity (Phosphomolybdenum antioxidant assay)** The method described by Prieto *et al.* (1999) was used to determine the ability of HLML, HLMR and LOML to reduce Mo-VI to Mo-V with subsequent formation of green phosphate-molybdate complex in an acidic pH condition. HLML, HLMR, LOML and standard drug (α-tocopherol) concentrations ranging from 0.5 to 10 mg/mL and 0.008 to 0.03 mg/mL respectively were prepared. To 1 mL each of HLML, HLMR, LOML and standard drug, 3 mL of reagent solution (0.6 M sulphuric acid, 28 mM disodium phosphate and 4 mM ammonium molybdate) was added and incubated at 95<sup>o</sup>C for 90 mins. A blank (all reagents without extract/standard drug) was treated in the same manner. The mixture was allowed to cool and absorbance was measured at 695 nm.

#### 2.2.7 Experimental animals

Sprague-Dawley rats (150 to 200g) were obtained from the Department of Pharmacology, Faculty of Pharmacy and Pharmaceutical Sciences, Kwame Nkrumah University of Science and Technology (KNUST), Kumasi, Ghana animal house and kept in stainless steel cages with soft wood shavings as bedding. They were maintained under standard environmental conditions of temperature  $(30\pm2^{\circ}C)$  and adequate humidity, with a twelve hour cycle of light and darkness. The animals were fed with standard pellet diet and provided with water *ad libitum*.

#### 2.2.8 Ethical clearance

The experiments were conducted in accordance with accepted principles for laboratory animal use and care (EU directive of 1986:86/609/EEC) and approval from the Animal Ethical Committee, Faculty of Pharmacy and Pharmaceutical Sciences, KNUST, Kumasi, Ghana.

#### 2.2.9 Determination of acute anti-inflammatory activity of extracts

The carrageenan induced foot oedema model in rats as described by Winter *et al.* (1962) was employed to assess the extracts ability to inhibit/reduce the paw oedema or swelling. Two experiments were performed to assess the acute anti-inflammatory activity of HLML, HLMR, and LOML. The first was to determine the effect of the drugs given 1 h before inducing the oedema (prophylaxis), and the second was to examine the therapeutic effect of the drug, whereby the drug was administered 1 h post oedema induction.

Sprague-Dawley rats were weighed and assigned randomly into groups consisting of four rats each. The initial foot volumes of the rats were measured using an electronic caliper after which reference drug (100 mg/kg aspirin) and extracts (30, 100 and 300

mg/kg), prepared in sterile distilled water, administered orally to the respective groups, with the control group given 0.5 mL sterile distilled water. After an hour of administering the drugs, oedema was induced by injecting 0.1 ml of 1% w/v carrageenan solution (in sterile distilled water) into the sub-plantar tissue of the right foot pads of the rats. Foot volumes were again measured at an hourly interval for 6 hours.

In the therapeutic experimental protocol, eodema was induced and measured 1h before administering the extracts and reference drug.

Inhibition of inflammation was calculated using the relation;

% change in paw thickness =  $100 \text{ x} \left[ \frac{(T_f - T_i)}{T_i} \right]$ Where

T<sub>i</sub> is paw thickness before carrageenan injection

T<sub>f</sub> is paw thickness at time T.

Raw scores for right foot thickness were individually normalised as percentage of change from their values at time 0 and then averaged. Total pedal oedema was calculated in arbitrary units as the area under the curve (AUC) and to determine the percentage inhibition of oedema, the following equation was used:

% inhibition of oedema = 
$$\left[\frac{AUC_{(control)} - AUC_{(treatment)}}{AUC_{(control)}}\right] \times 100$$

#### 2.2.10 Wound healing activity of extracts

The ability of HLML, HLMR and LOML to promote wound healing *in vivo* was assessed using the excision wound healing model.

#### 2.2.10.1 Formulation of extracts into cream

Aqueous cream base was prepared following the protocol described in the British Pharmacopoeia (2000), without the preservative (phenoxyethanol). HLML, HLMR and LOML were incorporated into the aqueous cream to give different concentrations of 5 and 10% w/w. The formulated creams were stored in the refrigerator at 25°C and constantly monitored for physical changes like phase separation, colour, odour, and texture.

#### 2.2.10.2 Excision wound model

The excision wound model described by Morton and Malone (1972) and modified by Agyare *et al.* (2013c) was employed in this study. The dorsal fur of rats was completely removed using new and clean razor blades. The area that was shaved was neatly wiped with 70% v/v ethanol and the rats were anaesthetized with an intramuscular injection of ketamine hydrochloride (50 mg/kg body weight). Sterile surgical scissors and toothed forceps were used to make full thickness wounds of about 20 to 25 mm diameter at the back of the rats.

The rats were randomly selected into different groups with five (5) rats in each group. Each group received a respective treatment as follows:

Group A: No treatment

Group B: Treated with aqueous cream base

Group C: Treated with silver sulphadiazine (1% w/w)

Group D and E: Received 5 and 10% w/w HLML aqueous creams, respectively. Group F and G: Received 5 and 10% w/w HLMR aqueous creams, respectively.

Group H and I: Received 5 and 10% w/w LOML aqueous creams, respectively.

#### 2.2.10.3 Treatment of wounds and wound diameter assessment

Wounds were not treated for the first 24 h period, after which they were treated daily for fifteen (15) days by topical application of 0.1 g of the respective creams. All the wounds were cleansed with normal saline (0.9% w/v) before treatment.

Diameter of the excised wounds was measured with the aid of a pair of divider and meter rule on every other day to assess the rate of wound contraction. Percentage wound closure was calculated as a percentage of the original wound from day 1 post wounding and result expressed as mean  $\pm$  standard error mean (SEM). The equation below was used to calculate the percentage wound closure:

% wound closure = [(wound size <sub>initial</sub> - wound size <sub>final</sub>)/ wound size <sub>initial</sub>]  $\times$  100

# 2.2.10.4 Histological studies

Treated and untreated wound tissues from all the groups were excised and kept in 10% neutral buffered formalin on the 15<sup>th</sup> day post wounding for histological examination to assess the degree of tissue repair (re-epithelisation, angiogenesis, collagenation and granular tissue formation). The tissues stored in buffered formalin were dehydrated several times in different concentrations of ethanol (50%, 70%, 95% and 100%), after which they were rinsed a number of times in xylene to remove ethanol and then embedded in paraffin to strengthen them for easy dissection. These tissues were sectioned into a thickness of about 5  $\mu$ m and later 'deparaffinised'. The tissues were then mounted on a clean glass slides and stained with haematoxylin and eosin, they were finally observed under the microscope (Udupa *et al.*, 1995; Chung *et al.*, 2010; Talekar *et al.*, 2012).

#### 2.2.11 In vitro cell toxicity studies of extracts

The effect of the extracts on cell viability and LDH release was investigated using HaCaT-keratinocytes (skin cell line) which was kindly provided by Professor N. E. Fusenig, German Cancer Research Center (Deutsche Krebsforschungszentrum) in Heidelberg, Germany.

#### 2.2.11.1 Determination of the influence of HLML, HLMR and LOML on cell viability

The method as described by Mosmann (1983) and modified by Agyare *et al.* (2011) was employed to assess the effect of HLML, HLMR and LOML on cell viability. Different concentrations of HLML, HLMR, and LOML (0.1, 1.0, 10.0 and 100.0  $\mu$ g/mL) were prepared in HaCaT keratinocytes medium in 96-well micro-titre plate and seeded with 10<sup>5</sup> cells per well. The plates were then incubated in 5% CO<sub>2</sub> at 35°C after which the medium was removed and cells washed with 100  $\mu$ L of phosphate buffer solution (PBS). This was followed by the addition of 50  $\mu$ L MTT solution and reincubated to allow the formation of insoluble formazan crystals, which were then dissolved in 50  $\mu$ L of DMSO to give a violet colour. The intensity of the violet coloured dissolved formazan crystals was measured by taking absorbance at 595 nm against 690nm, which correlates with the number of viable cells. HaCaT keratinocyte cells treated with 1% foetal calf serum (FCS) and untreated cells served as positive and negative controls respectively.

#### 2.2.11.2 LDH cell toxicity assay

HaCaT keratinocytes ( $10^5$  cells per well) were incubated at  $35^{\circ}$ C for 24 h in the presence of different concentrations (0.1, 1.0, 10.0 and 100.0 µg/mL) of HLML, HLMR and LOML. Twenty-five micro-litres of the reaction mixture (Lactate, NAD+, INT and diaphorase) was then added and incubated in the dark at 20°C for 30 min. The reaction was terminated by the addition of 10 µL 1M HCL solution to each well. The absorbance of the resultant solutions was measured with MTP reader at 490 against 690 nm. Untreated cells and cells treated with lysis solution (10% Triton X100 in 5% FCS) were used as negative and positive controls respectively. (Decker and

Lohmann-Matthes, 1988)

# 2.3 Statistical analysis

Data were presented as the mean ± standard error mean (SEM) in the studies. Analysis of results was done using one-way and two-way ANOVA followed by the Dunnett's/Tukey's and Bonferroni's *post hoc* test respectively. Graphs were plotted with Graph Pad Prism for windows version 6 (Graph Pad, San Diego, CA, USA).



#### **CHAPTER THREE**

# **3.0 RESULTS**

#### 3.1 Plant extraction

Extraction of medicinal plant parts with a selective solvent is necessary to attain desired therapeutic portions and to exclude unwanted material (Tiwari *et al.*, 2011). The maceration extraction procedure is highly accepted since most plant constituents may be thermolabile (Ncube *et al.*, 2008). The quantity of the 70% methanol extract obtained after cold maceration was expressed as percentage in relation to the quantity of the dried powdered plant materials used for the extraction, to give the percentage yield. The percentage yield for HLML, HLMR and LOML were 17.49, 7.50 and

11.29%, respectively (Table 3.1.1).

Table 3.1.1 Percentage yield of methanol extracts of *H. latifolia* and *L. ovalifolia* 

Code	Extracts	% yield
HLML	Hilleria latifolia leaf methanol extract	17.49
HLMR	Hilleria latifolia root methanol extract	7.50
LOML	Laportea ovalifolia leaf methanol extract	11.29

# **3.2 Preliminary phytochemical screening**

Phytochemical screening has been used over the years to detect diverse groups of naturally occurring plant metabolites and is considered as an effective approach for discovering bioactive profile of plants of therapeutic importance (Masih and Singh, 2012). Various phytochemical tests were performed on HLML, HLMR and LOML and their dried powdered plant materials, respectively to identify their phytochemical composition. Phytochemical screening of HLML and the dried powdered leaf material of *H. latifolia* revealed the presence of tannins, flavonoids, glycosides, saponins, alkaloids, sterols and terpenoids. In HLMR and the dried powdered root material of *H. latifolia*, tannins, glycosides, saponins, alkaloids, sterols and terpenoids were present whereas flavonoids were absent. LOML and the pulverized leaf material of *L. ovalifolia* also showed the presence of tannins, glycosides, sterols and terpenoids. Additionally, saponins were also found in the pulverized leaf material of *L. ovalifolia* (Table 3.2.1).

 Table 3.2.1: Phytochemical screening of HLML, HLMR, LOML and their dried powdered plant samples

Secondary	H. latifolia leaf		L. ovalif	olia leaf	H. latifolia root	
metabolites	HLML	Powdered	LOML	Powdered	HLMR	Powdered
		plant material	-	plant material		plant material
Tannins	+	+	+	+	+	+
Flavonoids	+	+	-	-		-
Glycosides	+	+	+	+	1-+-	+
Saponin	+	+	7 62	+	2+	+
Alkaloids	+	+	11	Dia	+	+
Sterols	+	+	< +	+	+	+
Terpenoids	+	+	+		+	+

(+) = presence of secondary metabolites; (-) = absence of secondary metabolites. HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanol extract, HLMR: *Hilleria latifolia* root methanol extract.

# **3.3 HPLC Profiles of extracts**

The HPLC profiles of HLML, HLMR and LOML were developed at wavelength of 254 nm, using a mobile phase of 20% methanol, flow rate of 1 mL/min and pressure of 15 MPa. The chemical profiling of HLML, HLMR and LOML shown below indicates the metabolites present at the wavelength used and will serve as a guide in identifying the plants.







Figure 3.3.2: HPLC profile of HLMR at 254 nm.



Figure 3.3.3: HPLC profile of LOML at 254 nm.

#### 3.4 Antimicrobial activity of extracts

The broth dilution method is reproducible and does not rely on the ability of the test antimicrobial agent to diffuse through culture media making it suitable method for determination of MIC of extracts (Cos *et al.*, 2006). Hence the broth dilution method was employed to determine the antimicrobial activity and MIC of HLML, HLMR and LOML.

The MICs of HLML, HLMR and LOML against typed and clinical strains of microorganisms, consisting of Gram-positive bacteria (*B. subtilis*, *S. aureus*, *E. feacalis*, *S. pyogenes*), Gram negative bacteria (*E. coli*, *S. typhi*, *K. pneumonia*, *P. aeruginosa*) and fungus (*C. albicans*) were 50 and100 mg/mL. Whereas that of ciprofloxacin was 2 to 4  $\mu$ g/mL against test bacteria and ketoconazole was 10  $\mu$ g/mL against test fungi (Table 3.4.1).

Test	MIC of extract	MIC of	reference		
microorganisms		antibiotics (µg/mL)			
	HLML	HLMR	LOML	Cipro	Keto
E. coli	50.0	50.0	100.0	2.0	nd
B. subtilis	50.0	50.0	50.0	2.0	nd
S. typhi	50.0	50.0	50.0	4.0	nd
K. pneu <mark>moniae</mark>	50.0	50.0	50.0	4.0	nd
S. aureus	50.0	50.0	100.0	4.0	nd
E. feacalis	50.0	100.0	100.0	4.0	nd
S. pyogenes	50.0	50.0	100.0	4.0	nd
P. aeruginosa	50.0	50.0	50.0	4.0	nd
C. albicans	50.0	100.0	100.0	nd	10.0

Table 3.4.1 MIC of HLML, HLMR and LOML against test organisms

nd = not determined, Cipro = Ciprofloxacin, Keto = ketoconazole. Typed strains (*E. coli*, *B. subtilis*, *S. typhi*, *S. aureus*, *E. feacalis*, *P. aeruginosa*), clinical strains (*K. pneumonia*, *S. pyogenes*, *C. albicans*). HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanol extract, HLMR: *Hilleria latifolia* root methanol extract.

#### 3.5 Antibiotic modifying activity by HLML and LOML

HLML and LOML ability to modify the activity of some selected antibiotics were evaluated using the broth dilution method with few modifications as described by Adu *et al.* (2014). The MICs of the antibiotics (amoxicillin, erythromycin, ciprofloxacin, tetracycline and ampicillin) were first determined, after which the MICs of the antibiotics were determined again in the presence of sub-inhibitory concentration (5 mg/mL) of HLML and LOML. The activity of amoxicillin against *E. coli, B. subtilis, S. typhi, S. aureus* and *P. aeruginosa* were potentiated in the presence of the subinhibitory concentration of HLML. The sub-inhibitory concentration of HLML again enhanced the activity of ampicillin against *E. coli* and *S. typhi,* as well as tetracycline

against *K. pneumonia*. However, both HLML and LOML sub-inhibitory concentrations reduced the activities of erythromycin and ciprofloxacin against all test organisms (Table 3.5.1).

some une	biotics					Contraction of the local division of the loc				
Antibiotic	cs only/		-		Test 1	nicroorga	anisms			
HLML/L	OML.	EC	BS	ST	KP	SA	EF	SP	PA	
		Number of folds increase in activity								
Amoxy	HLML	8	4	2	1	8	alo	>0.25	2	
Amoxy	LOML	2	H.	0.5	0.5	2	1	>0.25	1	
Ampi	HLML	2	1	4	1	1	1	>0.0625	1	
1	LOML	1	0.25	0.25	0.5	1	1	>0.0625	>0.0625	
	HLML	0.5	1	1	4	1	1	1	2	

Table 3.5.1: Effect of sub-inhibitory concentration (5mg/mL) of HLML and LOML on some antibiotics

Tetra									
	LOML	0.125	0.125	0.25	0.5	0.25	0.25	0.25	0.5
	HLML	0.0625	0.25	0.0625	0.125	0.0625	0.125	0.125	0.25
Erythro	LOML	0.0625	0.5	0.125	0.125	0.125	0.25	0.0625	0.125
Cipro	HLML	0.125	0.125	0.25	0.5	0.5	0.5	0.5	0.25
Cipio	LOML	0.125	0.0625	0.125	0.125	0.125	0.25	0.25	0.125

*EC: E. coli, BS: B. subtilis, ST: S. typhi, KP: K. pneumonia, SA: S. aureus, EF: E. feacalis, SP: S. pyogenes, PA: P. aeruginosa.* Amoxy-amoxicillin, Ampi-ampicillin, Tetra-tetracycline, Erythro-erythromycin, Cipro-ciprofloxacin, HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanol extract, HLMR: *Hilleria latifolia* root methanol extract.

# 3.6 Antioxidant activity of extracts

# 3.6.1 DPPH free radical scavenging activity of HLML, HLMR and LOML

HLML, HLMR and LOML ability to scavenge DPPH free radicals was studied using the method described by Prior *et al.* (2003). The method is very simple and rapid and is based on transfer of electrons or hydrogen atom by an antioxidant agent to reduce DPPH (Huang *et al.*, 2005; Badarinath *et al.*, 2010). The reduced DPPH is observed as colour change from violet to yellow, which was measured spectrophotometrically at 517 nm.

HLML, LOML, HLMR and vitamin E ( $\alpha$ -tocopherol) showed antioxidant activity at the test concentrations (0.0156 to 1 mg/mL). The IC<sub>50</sub> values obtained for vitamin E, HLML, HLMR and LOML were 18.9 ± 1.3, 102.5 ± 1.5, 233.5 ± 0.5 and 130.8 ± 0.9 µg/mL, respectively (Figure 3.6.1 .2).



**Figure 3.6.1.1**: DPPH free radical scavenging activity of HLML, LOML, HLMR and standard drug (Vitamin E). conc: concentration, Vit E: Vitamin E, HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanol extract, HLMR: *Hilleria latifolia* root methanol extract. Values are mean ± SEM.

# Table 3.6.1.1: IC<sub>50</sub> of extracts and vitamin E (α-tocopherol)

Extract/ reference compound	HLML	LOML	HLMR	vitamin E
	me			1

 $IC_{50} (\mu g/mL)$  $102.5 \pm 1.477$  $130.8 \pm 0.8511$  $233.5 \pm 0.4933$  $18.90 \pm 1.312$ HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanolextract, HLMR: *Hilleria latifolia* root methanol extract. Values are mean  $\pm$  SEM.

# 3.6.2 Total phenol content of HLML, HLMR and LOML

Folin-Ciocalteu reagent was used to assess the total phenolic content of HLML, HLMR and LOML, as described by Škerget *et al.* (2005). Phenolic compounds have the ability to reduce Folin-Ciocalteu reagent in a basic environment or medium to form a blue complex, and this can be quantified spectrophotometrically at 760 nm (Socha *et al.*,

2009). The total phenol content was calculated as milligram tannic acid equivalence (TAE) per gram of the extract using the tannic acid calibration or standard curve.

HLML, HLMR and LOML at the test concentrations (0.5 to 10 mg/mL) showed an increased phenol content with increasing concentration. HLML ranged from 10 to198 mg TAE/g of the extract, HLMR was from 27 to 175 mg TAE/g of the extract and LOML stretched from 7 to 122 mg TAE/g of the extract (Figures 3.6.2.1, 3.6.2.2 and 3.6.2.3). The total phenol content in each of the extracts (HLML, HLMR and LOML) was calculated as mean  $\pm$  SEM as indicated in table 3.6.2.1, with HLML having the highest total phenol content of 103.0  $\pm$  1.335 mg/g.



**Figure 3.6.2.1:** Total phenol content in different concentrations of HLML expressed as tannic acid equivalent. HLML: *Hilleria latifolia* leaf methanol extract. Values are means  $\pm$  SEM \*\*\*\**p*<0.0001; \*\*\**p*< 0.001; \*\*\**p*< 0.01 (Tukey's *post hoc* test).



**Figure 3.6.2.2:** Total phenol content in different concentrations of HLMR expressed as tannic acid equivalent. HLMR: *Hilleria latifolia* root methanol extract. Values are means  $\pm$  SEM. \*\*\*p< 0.001 (Tukey's *post hoc* test).



**Figure 3.6.2.3:** Total phenol content in different concentrations of LOML expressed as tannic acid equivalent. LOML: *Laportea ovalifolia* leaf methanol extract. Values are means  $\pm$  SEM. \*\*\*\*p<0.0001; \*\*p<0.01 (Tukey's *post hoc* test).

# Table 3.6.2.1 Total phenol content of HLML, HLMR and LOML
Extracts	HLML	LOML	HLMR
Total phenol content (mg	$103.0\pm1.335$	$56.75\pm0.3220$	$91.32\pm4.258$
TAE/g extract)			

TAE: Tannic acid equivalent, HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanol extract, HLMR: *Hilleria latifolia* root methanol extract. Values are mean  $\pm$  SEM.

# 3.6.3 Total antioxidant capacity of HLML, HLMR and LOML

The phosphomolybdenum antioxidant assay as described by Prieto et al. (1999) and

Amponsah et al. (2014) was employed to determine the total antioxidant capacity of

HLML, HLMR and LOML, using  $\alpha$ -tocopherol as standard drug. In an acidic

condition, an antioxidant compound is able to reduce  $Mo^{+6}$  to  $Mo^{+5}$  with subsequent formation of a green complex which can be quantified by measuring absorbance at 695nm. Total antioxidant capacity of HLML, HLMR and LOML was expressed as atocopherol equivalence ( $\alpha$ -TE) from the  $\alpha$ -tocopherol calibration curve.

The total antioxidant capacity of HLML was  $410.4 \pm 4.732 \text{ mg} \alpha$ -TE/g extract, HLMR was  $408.0 \pm 18.70 \text{ mg} \alpha$ -TE/g and that of LOML was calculated to be  $337.6 \pm 6.961 \text{ mg} \alpha$ -TE/g extract (Table 3.6.3.1). Values are presented as mean  $\pm$  standard error mean. At the test concentrations (0.5 to 10 mg/mL) of HLML, HLMR and LOML, it was observed that, antioxidant capacity decreased as the concentrations reduced, (Figures 3.6.3.1, 3.6.3.2 and 3.6.3.3).

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**Figure 3.6.3.1**: Total antioxidant capacity in different concentrations of HLML expressed as  $\alpha$ tocopherol equivalent. HLML: *Hilleria latifolia* leaf methanol extract. Values are means  $\pm$ SEM. \*\*\*\**p*<0.0001; \*\**p*< 0.01 (Tukey's *post hoc* test).



**Figure 3.6.3.2:** Total antioxidant capacity in different concentrations of HLMR expressed as  $\alpha$ -tocopherol equivalent. HLMR: *Hilleria latifolia* root methanol extract. Values are means  $\pm$  SEM. (Tukey's *post hoc* test).



**Figure 3.6.3.3:** Total antioxidant capacity in different concentrations of LOML expressed as  $\alpha$ tocopherol equivalent. LOML: *Laportea ovalifolia* leaf methanol extract. Values are means  $\pm$ SEM. \*\*p< 0.01 (Tukey's *post hoc* test).

Table 3.6.3.1: Total antioxidant capacity of H	LIVIL,	HLMK	and LOMI
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Extracts	HLML	LOML	HLMR
Total antioxidant capacity	410.4 ± 4.732	337.6 ± 6.961	$408.0 \pm 18.70$
(mg $\alpha$ -TE/g extract)			

 $\alpha$ -TE:  $\alpha$ -tocopherol equivalent, HLML: *Hilleria latifolia* leaf methanol extract, LOML: *Laportea ovalifolia* leaf methanol extract, HLMR: *Hilleria latifolia* root methanol extract. Values are mean  $\pm$  SEM.

## 3.7 Acute anti-inflammatory activity of HLML, HLMR and LOML

The preventive and curative effects of HLML, HLMR and LOML were examined using the carrageenan-induced foot oedema model in rats (Winter *et al.*, 1962). This model is acceptable and most widely used animal model of acute inflammation. Carrageenan has the ability to induce an intense and reproducible inflammation by causing the release of more than one inflammatory mediator which is very useful in assessing an agent's antiinflammatory activity (Winter *et al.*, 1962; Kaur *et al.*, 2004). Rats were grouped randomly with each group consisting of four (4) animals.

The groups were treated as follows 30, 100 and 300 mg/kg of HLML, HLMR and LOML, 100 mg/kg of aspirin (reference drug) and control group given 0.5 mL sterile distilled water (vehicle) before (preventive) or after (curative) induction of oedema.

HLML (30, 100, 300 mg/Kg) when administered before (preventive) the induction of the carrageenan paw oedema caused the mean maximal swelling attained at 2 h to be significantly (p< 0.01) reduced to 41.81 ± 3.14%, 51.21 ± 5.79% and 38.13 ± 3.21%, respectively from the inflamed control response of 61.50 ± 1.44% (Figure 3.7.1A). The total paw swellings induced over the 6 h (measured as the area under the time course curve, AUC) were also significantly (p<0.001) suppressed to 64.12 ± 4.88%, and 55.70 ± 6.14% of the inflamed control response for 30 and 300 mg/Kg (Figure

3.7.1B). Also, HLMR (30, 100 and 300 mg/Kg) showed a significant (p<0.001) reduction of the oedema with the mean maximal swelling attained at 2 h reduced to 54.04 ± 1.40%, 48.33 ± 0.54% and 41.30 ± 2.36% respectively, from the inflamed control response of 61.50 ± 1.44% (Figure 3.7.2A). The total paw swellings induced over the 6 h were also significantly (p<0.001) suppressed to 88.51 ± 4.30%, 73.38 ± 2.99% and 61.78 ± 3.64% respectively (Figure 3.7.2B). Similarly, LOML (30, 100 and 300 mg/Kg) showed a significant (p<0.001) reduction of the oedema with the mean maximal swelling attained at 2 h reduced respectively to 55.38 ± 0.57%, 51.15 ± 2.30% and 45.05 ± 2.81% from the inflamed control response of 61.50 ± 1.44% (Figure 3.7.3A). The total paw swellings induced over the 6 h were also significantly (p<0.001) reduction of the oedema with the mean maximal swelling attained at 2 h reduced respectively to 55.38 ± 0.57%, 51.15 ± 2.30% and 45.05 ± 2.81% from the inflamed control response of 61.50 ± 1.44% (Figure 3.7.3A). The total paw swellings induced over the 6 h were also significantly (p<0.001) suppressed to 86.96 ± 2.14%, 76.37 ± 3.70% and 67.92 ± 3.77% respectively of the inflamed control response (Figure 3.7.3B).

When administered after the induction of the carrageenan paw oedema (curative), HLML (30, 100, 300 mg/Kg) suppressed significantly (p<0.001) the mean maximal swelling attained at 4 h respectively to  $61.96 \pm 7.393\%$ ,  $58.66 \pm 1.852\%$  and  $69.62 \pm$ 2.729% of the inflamed control response of  $105.2 \pm 4.553\%$  (Figure 3.7.4A). The total paw swellings induced over the 6 h was significantly (p<0.001) suppressed to  $64.63 \pm$ 8.39%,  $59.52 \pm 3.46\%$  and  $69.92 \pm 5.39\%$ , respectively of the inflamed control response (Figure 3.7.4B). Similarly, HLMR (30, 100, 300 mg/Kg) showed a significant (p<0.001) inhibition of the oedema with the mean maximal swelling at 4 h reduced respectively to  $63.70 \pm 5.490\%$ ,  $75.22 \pm 7.452\%$  and  $65.70 \pm 4.943\%$  of the inflamed control response of  $105.2 \pm 4.553\%$  (Figure 3.7.5A). The total paw swellings induced over the 6 h were also significantly (p<0.001) suppressed to  $64.30 \pm 4.66\%$ ,  $75.05 \pm$ 9.59% and  $62.00 \pm 4.90\%$  respectively of the inflamed control response (Figure 3.7.5B). Again, LOML (30, 100, 300 mg/Kg) showed a significant

(p<0.001) inhibition of the oedema with the mean maximal swelling at 4 h reduced respectively to  $45.51 \pm 0.6827\%$ ,  $43.91 \pm 1.431\%$  and  $38.19 \pm 0.9960\%$  of the inflamed control response of  $105.2 \pm 4.553\%$  (Figure 3.7.6A). The total paw swellings induced over the 6 h were also significantly (p<0.001) suppressed to  $48.04 \pm 1.37\%$ ,  $45.38 \pm 1.57\%$  and  $40.79 \pm 2.40\%$  respectively of the inflamed control response (Figure 3.7.6B).

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**Figure 3.7.1:** Prophylactic effect of HLML on carrageenan-induced oedema in rats. A: Timecourse curve, **B:** AUC of carrageenan induced oedema, Ctrl: Control group, Asp: Aspirin treated group, HLML: *Hilleria latifolia* leaf methanol extract. Values are means  $\pm$  SEM (n=4). \*\*\*p<0.001; \*\*p< 0.01 (Dunnett's *post hoc* test).



**Figure 3.7.2:** Prophylactic effect of HLMR on carrageenan-induced oedema in rats. A: Timecourse curve, **B**: AUC of carrageenan induced oedema, Ctrl: Control group, Asp: Aspirin treated group, HLMR: *Hilleria latifolia* root methanol extract. Values are means  $\pm$  SEM (n=4) \*\*\*p<0.001; \*p<0.05 compared to control group (Dunnett's *post hoc* test).

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**Figure 3.7.3:** Prophylactic effect of LOML on carrageenan-induced oedema in rats. A: Timecourse curve, **B**: AUC of carrageenan induced oedema, Ctrl: Control group, Asp: Aspirin treated group, LOML: *Laportea ovalifolia* leaf methanol extract. Values are means  $\pm$  SEM (n=4) \*\*\**p*<0.001; \*\**p*<0.01 compared to control group (Dunnett's *post hoc* test).



**Figure 3.7.4:** Therapeutic effect of HLML on carrageenan-induced oedema in rats. A: Timecourse curve, **B**: AUC of carrageenan induced oedema, HLML: *Hilleria latifolia* leaf methanol extract. Values are means  $\pm$  SEM (n=4) \*\*\**p*<0.001; compared to control group (Dunnett's *post hoc* test). Arrow indicates point of Aspirin or HLML administration.



**Figure 3.7.5:** Therapeutic effect of HLMR on carrageenan-induced oedema in rats. A: Timecourse curve, **B**: AUC of carrageenan induced oedema, HLMR: *Hilleria latifolia* root methanol extract. Values are means  $\pm$  SEM (n=4) \*\*\*p<0.001; \*\*p<0.01 compared to control group (Dunnett's *post hoc* test). Arrow indicates point of Aspirin or HLMR administration.



**Figure 3.7.6:** Therapeutic effect of LOML on carrageenan-induced oedema in rats. **A:** Timecourse curve, **B:** AUC of carrageenan induced oedema, LOML: *Laportea ovalifolia* leaf methanol extract. Values are means  $\pm$  SEM (n=4) \*\*\*p<0.001 compared to control group (Dunnett's *post hoc* test). Arrow indicates point of Aspirin or LOML administration.

#### 3.8 Wound healing activity of extracts

The excision wound model as described by Agyare *et al.* (2013c) was used to assess the influence of HLML, HLMR and LOML on wound healing. HLML, HLMR and LOML were formulated into an aqueous cream of 5 and 10% w/w each and applied on wounds daily for fifteen (15) days. Silver sulphadiazine (1% w/w) was used as reference drug, vehicle group was treated with only aqueous cream. The rate of wound closure (wound diameter) was measured every other day through the period of treatment from the day of wound creation.

HLML, HLMR and LOML at 5 and 10% significantly (p<0.001) reduced wound size from day 5 to 15 after injury (Figures 3.8.1a, 3.8.2a and 3.8.3a). Also, their respective area under the curve (AUC) revealed a significant reduction (p<0.001) in wound size when compared to the untreated group (Figures 3.8.1b, 3.8.2b and 3.8.3b).



**Figure 3.8.1:** Influence of HLML on wound closure. **A:** time-course curve, **B:** AUC of time course curve, SS: Silver sulphadiazine, HLML: *Hilleria latifolia* leaf methanol extract. Values are mean  $\pm$ SEM (n=5). \*\*\**p*<0.001; \*\**p*<0.05 compared to untreated group (One-way and Two-way ANOVA followed by Dunnett's and Bonferroni's *post hoc* tests, respectively).



**Figure 3.8.2:** Influence of HLMR on wound closure. **A:** time-course curve, **B:** AUC of time course curve, SS: Silver sulphadiazine HLMR: *Hilleria latifolia* root methanol extract. Values are mean  $\pm$ SEM (n=5). \*\*\*p<0.001; \*\*p<0.01; \*p<0.05 compared to untreated group (Oneway and Two-way ANOVA followed by Dunnett's and Bonferroni's *post hoc* tests, respectively).



**Figure 3.8.3:** Influence of LOML on wound closure. **A:** time-course curve, **B:** AUC of time course curve, SS: Silver sulphadiazine, LOML: *Laportea ovalifolia* leaf methanol extract. Values are mean  $\pm$  SEM (n=5). \*\*\*p<0.001; \*\*p<0.01; \*p<0.05 compared to untreated group

(One-way and Two-way ANOVA followed by Dunnett's and Bonferroni's *post hoc* tests, respectively).

**3.9 Histological studies of wound tissues treated with HLML, HLMR and LOML** On the 15<sup>th</sup> day post wounding, wounds from all the groups were excised for histological examination to microscopically identify and assess morphological changes in tissues such as re-epithelisation, angiogenesis, collagenation and granular tissue formation. Heamatoxylin and eosin (H&E) staining was used for the study since the staining process is simple and also serves as a general purpose differential staining technique for differentiating between nuclear bodies and other organelles such as collagen, muscles and connective fibres in the cytoplasm (Udupa *et al.*, 1995; Chung *et al.*, 2010).

The images from the histological study revealed the following morphological characteristics:

- Untreated wound tissues revealed persistent inflammation with marked tissue necrosis and little proliferation indicating poor wound healing rate.
- Wound tissues treated with aqueous cream only (Vehicle) revealed presence of persistent inflammation and little proliferation which indicated poor wound healing rate.
- Silver sulphadiazine (1% w/w)-treated wound tissues were characterised with granulation tissue formation and angiogenesis with evidence of fibroblast proliferation and collagen synthesis indicating re-epithelialisation and wound healing.
- HLML (10%)-treated wound tissues had increased collagenation and reepithelialisation, which indicates high rate of wound healing, while HLML

(5%)-treated wound tissue showed appreciable angiogenesis and fibroblastic

activity with evidence of substantial collagen deposition in wounds (Figure 3.9.1).

- HLMR (10%) treated wound tissue was marked with an increase collagenation and re-epithelisation indicative of high rate of wound healing, whereas appreciable angiogenesis and fibroblastic activity with evidence of considerable collagen deposition were observed in wound tissue treated with 5% HLMR (Figure 3.9.2).
- Wound tissue treated with LOML (10%) showed high rate of wound healing with increased collagenation and re-epithelisation, while tissue treated with LOML (5%) indicated appreciable angiogenesis and fibroblastic activity with evidence of considerable collagen deposition in wounds (Figure 3.9.3).





**Figure 3.9.1:** Histological images (x400) of wound tissues from HLML, LOML and HLMRtreated and untreated wounds. **A**: Untreated wound tissues, **B**: vehicle-treated (aqueous cream only) wound tissues, **C**: 1% w/w siver sulphadiazine-treated wound tissues, **D**: 10% w/w HLML-treated wound tissues, **E**: 5% w/w HLML-treated wound tissues, **F**: 10% w/w LOMLtreated wound tissues, **G**: 5% w/w LOML-treated wound tissues, **H**: 10% w/w HLMR-treated wound tissues, **I**: 5% w/w HLMR-treated wound tissues, **D**: 10% w/w HLMR-treated wound tissues, **I**: 5% w/w HLMR-treated wound tissues, **D**: 10% w/w HLMR-treated wound tissues, **I**: 5% w/w HLMR-treated wound tissues, **D**: 10% w/w HLMR-treated wound tissues, **I**: 5% w/w HLMR-treated wound tissues, **D**: 10% w/w HLMR-treated wound ti

fibrous tissue, ASCKE: Atrophic squamous cell keratinized epithelium, SGT: Scanty granulation tissue, SeG: Sebaceous gland, SwG: Sweat gland.

## 3.10 Cell toxicity studies of extracts

## 3.10.1 Influence of HLML, HLMR and LOML on cell viability

The influence of HLML, HLMR and LOML on the viability of HaCaT-keratinocytes was investigated using the MTT assay as described by Agyare et al. (2011) and Lupu and Popescu (2013). Dehydrogenase enzymes in the mitochondria of metabolically active cells reduced yellow MTT to purple formazan, which was then quantified spectrophotometrically at 595 nm as a measure of viable cell (Niles *et al.*, 2008). HaCaT-ketainocytes were treated with different concentrations (0.1, 1.0, 10.0 and 100.0  $\mu$ g/mL) of HLML, HLMR and LOML to assess their influence on the viability of the cells.

Treatment with 100  $\mu$ g/mL of HLML (p<0.001) and HLMR (p<0.05) significantly reduced the viability of HaCaT-keratinocytes compared to the untreated cells. However, at the lower test concentrations (0.1, 1.0 and 10.0) and LOML-treated cells, there was no significant (p>0.05) difference when compared to the HaCaTKeratinocyte untreated cells, whereas the cells treated with 1% foetal calf serum (positive control) significantly (p < 0.001) increased the cells viability compared to the untreated cells (Figures 3.10.1.1) THIS AD SANE to 3.10.1.3).

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**Figure 3.10.1.1**: Influence of HLML on HaCaT-keratinocytes viability. uc: untreated cells; **pc**: positive control (1% foetal calf serum); HLML: *Hilleria latifolia* leaf methanol extract. Values are mean ±SEM. \*\*\*p<0.001.



**Figure 3.10.1.2:** Influence of HLMR on HaCaT-keratinocytes viability. uc: untreated cells; **pc**: positive control (1% foetal calf serum); HLMR: *Hilleria latifolia* root methanol extract. Values are mean  $\pm$ SEM. \*\*\*p<0.001; \*p<0.05.



**Figure 3.10.1.3:** Influence of LOML on HaCaT-keratinocytes viability. uc: untreated cells; **pc**: positive control (1% foetal calf serum); LOML: *Laportea ovalifolia* leaf methanol extract. Values are mean ±SEM. \*\*\*p<0.001.

#### 3.10.2 Influence of HLML, HLMR and LOML on LDH release

The influence of HLML, HLMR and LOML on the release of LDH from HaCaTkeratinocytes was assessed to measure the integrity of the cell membrane. LDH is a stable cytoplasmic enzyme present in all cells and is only released into an extracellular medium or cell culture supernatant upon damage of the plasma membrane, indicating cell damage or necrosis. The LDH released from cells in the presence of toxic agents or compounds oxidizes lactate to pyruvate which then reacts with INT (2-[4-Iodophenyl]-3-[4-nitrophenyl]-5-phenyltetrazoliumchloride) to form red coloured formazan. The formazan was soluble in water and was quantified spectrophotometrically at a wavelength of 490 against 690 nm, as a measure of the number of lysed or damaged cells (Decker and Lohmann-Matthes, 1988; Smith et al., 2011). HaCaT-keratinocytes were cultured in the presence of HLML, HLMR and

LOML (0.1, 1.0, 10.0 and 100.0 µg/mL) to assess their membrane integrity by measuring the level of LDH released. Untreated cell and cells lysed with 10% Triton X-100 in 5% FCS were used as negative and positive controls, respectively.

HLML, HLMR and LOML-treated cells at the test concentrations showed no significant (p>0.05) LDH release from HaCaT-keratinocytes compared to the untreated cells.



Figure 3.10.2.1: Influence of HLML on release of LDH from HaCaT-keratinocytes uc: untreated cells. LDH: Lactate dehydrogenase. HLML: Hilleria latifolia leaf methanol extract. Values are mean  $\pm$ SEM. \*\*\*p<0.001. NO BADW

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**Figure 3.10.2.2:** Influence of HLMR on release of LDH from HaCaT-keratinocytes **uc**: untreated cells. LDH: Lactate dehydrogenase. HLMR: *Hilleria latifolia* root methanol extract. Values are mean  $\pm$ SEM. \*\*\**p*<0.001.



**Figure 3.10.2.3:** Influence of LOML on release of LDH from HaCaT-keratinocytes. **uc**: untreated cells. LDH: Lactate dehydrogenase. LOML: *Laportea ovalifolia* leaf methanol extract. Values are mean  $\pm$ SEM. \*\*\*p<0.001.

#### **CHAPTER FOUR**

## **4.0 DISCUSSION**

Medicinal plants have been used over the years for the treatment of various ailments including infectious and metabolic (non-communicable) diseases (Ayyanar and Ignacimuthu, 2011; Calixto, 2005). Some of these plants have thoroughly been investigated and found to possess the pharmacological and biological activity for which they are traditionally used for (Agyare *et al.*, 2013c). There is the need to validate or otherwise the claims of medicinal plants including *Hilleria latifolia* and *Laportea ovalifolia* used traditionally for the treatment of infections and other diseases.

*H. latifolia* is used traditionally for the treatment of rheumatism, boils, wounds and some skin diseases, whereas *L. ovalifolia* is used to prevent excessive menstrual bleeding and for treatment of wounds (Agyare *et al.*, 2009; Mshana, 2000; Sofowora, 1996).

The preliminary phytochemical screening revealed the presence of tannins, glycosides, saponins, flavonoids, alkaloids, sterols and terpenoids in the leaves and root of H. *latifolia* similarly to what was reported by Abotsi *et al.* (2012) and Schmelzer and Gurib-Fakim, (2008). Tannins, glycosides, sterols and terpenoids were also present in *L. ovalifolia*, however, flavonoids and alkaloids were absent (Table 3.2.1). Essiett *et al.* (2011) reported the presence of phytochemicals such as tannins, glycosides, saponins, flavonoids and alkaloids. The absence of flavonoids and alkaloids in the leaves of *L. ovalifolia* may be due to the different geographical location of the plant, the season and time of collection which are all contributing factors leading to variations in the phytochemical constituents of plants of the same species (González\_Martínez *et al.*, 2006; Stackpole *et al.*, 2011).

Secondary metabolites such as tannins, flavonoids, glycosides, alkaloids, terpenoids and steroids present in the plants may be responsible for their pharmacological and biological properties (Sofowora, 1993; Barbosa-Filho *et al.*, 2006; Mbagwu *et al.*, 2006; Maganha *et al.*, 2010).

Secondary metabolites in plants act individually or in synergy to bring about a wide range of pharmacological and biological activities (Gurib-Fakim, 2006; JenkeKodama *et al.*, 2008; Kubmarawa *et al.*, 2008). Tannins (Buzzini *et al.*, 2008; Koleckar *et al.*, 2008; Fiori *et al.*, 2013), saponins (Gurib-Fakim, 2006; Murugan *et al.*, 2013), alkaloids (Gomes *et al.*, 2009; Singh and Kumar, 2012), flavonoids (GuribFakim, 2006; Xu and Lee, 2001), glycosides (Akdemir *et al.*, 2011) and terpenoids (Irfan *et al.*, 2014) have been reported to exhibit various pharmacological and biological activities including antimicrobial, antioxidant, anti-inflammatory, wound healing, anti-nociceptive, antiulcer, antiviral, antitumor and analgesic effects.

In addition to the phytochemical screening, HLPC profile of the 70% methanol extracts (HLML, HLMR and LOML) were developed for identification purposes. HPLC profiling is more specific and helps in easy identification and confirmation of plant on the basis of specific phytochemicals present. The profile again indicates the complex chemical composition of an extract and provides identification parameters to figure out alterations in formulated herbal products containing the extract (Tistaert *et al.*, 2012).

The HPLC profiles of the extracts (Figures 3.3.1 to 3.3.3) showed that the peaks representing compounds present in the extracts appeared in the early part (early elution) of the chromatogram. This observation may be due to the polar solvent (70% methanol) used for the extraction.

The antimicrobial activity and MICs of HLML, HLMR and LOML against test organisms were determined using the broth dilution method. HLML, HLMR and LOML exhibited a broad spectrum antimicrobial activity against *E. coli B. subtilis, S. typhi, K. pneumonia, S. aureus, E. feacalis, S. pyogenes, P. aeruginosa* and *C. albicans* with MIC ranging from 50 to 100 mg/mL (Table 3.4.1). The antimicrobial activity observed may be attributed to the phytochemical constituents present in the extracts (Nweze *et al.*, 2004; Edeoga *et al.*, 2005).

The high MIC or low antimicrobial activity of HLML, HLMR and LOML may be as a result of low amount of the active constituents in the extracts. This observation is similar to a reported by Okwulehie and Akanwa (2013) on antimicrobial activity of *L. ovalifolia*. In this study, *L. ovalifolia* at 50 mg/ml did not inhibit the growth of the test organisms. However, Assob *et al.* (2011) reported on the antimicrobial activity of *H. latifolia* with MIC of 0.6 to 2.5 mg/mL. The MIC of 50 to 100 mg/mL for *H. latifolia* observed in this study may be as a result of different extraction procedures used and different locality of the plant materials used which may lead to different composition in terms of primary and secondary metabolites. The MIC of HLML against all the test microorganisms was 50 mg/mL, while that of HLMR was 100 mg/mL against *E. feacalis* and *C. albicans*, and 50 mg/mL against the remaining test organisms.

Fabry *et al.*, (1998) and Navarro and Delgado, (1999) reported that plant extracts with MIC ranging between 2.5 and 8 mg/mL are likely to be potential source of strong antimicrobial compounds. This suggests that HLML, HLMR and LOML may not be potential source of effective antimicrobial compounds. It is also well noted that plant extracts with low antimicrobial activity may contain some phytoconstituents that can modify the antimicrobial activity of known antimicrobials especially against resistant bacteria (Adu *et al* 2009).

Sub-inhibitory concentration of 5 mg/mL HLML and LOML modified the activity of amoxicillin, erythromycin, ciprofloxacin, tetracycline and ampicillin by either potentiating or nullifying their activity against the test organisms. The sub-inhibitory concentration (5 mg/mL) of HLML and LOML enhanced the activity of amoxicillin against most of the test organisms. The activity of ampicillin and tetracycline against *E. coli*, *S. typhi* and *K. pneumonia*, *P. aeruginosa* respectively were potentiated in the presence of sub-inhibitory concentration of HLML. Erythromycin and ciprofloxacin lost their activities completely against all the test organisms (Table 3.5.1).

The increased activity of these antibiotics in the presence of the sub-inhibitory concentration may be attributed to the phytochemicals of the extracts. For instance, flavonoids have been reported to have the ability to reverse the resistance of *S. aureus* to some antibiotics (Aiyegoro *et al.*, 2009). Antimicrobials from plants, at subinhibitory concentrations, have been reported to be efficient in synergism with antibiotics by enhancing their antimicrobial actions (Kamatou *et al.*, 2006). The phytochemicals act by reversing the resistance mechanisms of some microorganisms, thereby rendering them susceptible to antibiotics (Tenover, 2006). Plants have also been known to produce multi-drug resistance inhibitors (MDRIs) to enhance the antimicrobial activities of compounds (Stermitz *et al.*, 2000).

The reduced or nullified activity of the antibiotics may be as a result of interactions between the phytochemicals in the extract and the antibiotics or the microorganisms. The phyto-constituents may react chemically with the antibiotics leading to loss of activity (Adu *et al.*, 2009; Adu *et al.*, 2014). It has been established that certain substances can shield microorganisms from the lethal effects of some antimicrobial agents (Keweloh *et al.*, 1989). For example, some phytochemicals can bind to the surface structures of microorganisms thereby reducing their permeability to antibiotics (Adu et al., 2014).

Furthermore, some of the phytocontituents may act as protein activators or coenzymes which bind to and activate enzymes responsible for resistance in an organism, making them resistant to a previously potent antibiotic (Lambert, 2002).

Antioxidant activity of HLML, HLMR and LOML was determined by assessing their DPPH free radical scavenging properties, total phenol content and total antioxidant capacity. The DPPH free radical scavenging activity of HLML, LOML and HLMR increased with increasing concentration (Figure 3.6.1.1).

The IC<sub>50</sub> gives an idea on the ability of an agent to mop up free radicals indicating its potency as an antioxidant. Lower IC<sub>50</sub> values indicate potent antioxidant activity (Apenteng *et al.*, 2014). The IC<sub>50</sub> values of the extracts (Table 3.6.1.1) indicate that HLML ( $102.5 \pm 1.5 \mu g/mL$ ) exhibited the highest antioxidant activity, followed by LOML ( $130.8 \pm 0.9 \mu g/mL$ ) and HLMR ( $233.5 \pm 0.5 \mu g/mL$ ) having the least antioxidant property.

The antioxidant activity observed in HLML, LOML and HLMR may be due to their phenolic and/or non-phenolic contents (Wu and Ng, 2008). Moreover, the highest antioxidant activity of HLML as reported by Abosti *et al.* (2012) may be due to the presence of flavonoids. Even though HLMR exhibited an antioxidant activity, it was not as potent as the HLML, since the IC<sub>50</sub> of HLMR was higher than that of HLML.

This may be due to absence of phenolic compounds like flavonoids in HLMR (Jain *et al.*, 2008; Belguidoum *et al.*, 2015).

For this reason, the total phenol content of HLML, LOML and HLMR was determined. From the result, HLML had the highest total phenol content of  $103.0 \pm 1.335$  mg/g, which may also contribute to its free radical scavenging activity. Redox properties of phenolic compounds play an important role in neutralizing free radicals, quenching singlet and triplet oxygen species, or decomposing peroxides (Rice-Evans *et al.*, 1996; Zheng and Wang, 2001). HLMR had higher total phenol content (91.32  $\pm$  4.258 mg/g) as compared to LOML of (56.75  $\pm$  0.3220 mg/g). This indicates that the higher free radical scavenging activity of LOML as compared to HLMR may be due to other non-phenolic compounds present (Conforti *et al.*, 2008).

The total antioxidant capacity of HLML, LOML and HLMR (410.4  $\pm$  4.732, 337.6  $\pm$  6.961 and 408.0  $\pm$  18.70 mg/g, respectively) correlated with their respective total phenol content. This indicates that the higher the total phenol content the greater the total antioxidant capacity of the extracts. This confirms the report that flavonoids and other phenolic compounds significantly cause the reduction of Mo<sup>+6</sup> to Mo<sup>+5</sup> (Khan *et al.*, 2012).

The acute anti-inflammatory activity of HLML, LOML and HLMR was determined using the carrageenan-induced oedema in rat foot model (Winter *et al.*, 1962). Carrageenan-induced oedema is a vascular response with exudation of inflammatory cells and fluids at the site of injury, caused by dilation of arterioles and eventual increase in permeability of post capillary venules (Vinegar *et al.*, 1987).

The molecular response to carrageenan-induced oedema is bi-phasic involving the release of diverse inflammatory mediators, characterised by marked oedema formation. Inflammatory mediators such as histamine, serotonin and bradykinin are released during the first phase (1 to 2 h), and sustained by the release of prostaglandins and nitric oxide in the second phase (Thomazzi *et al.*, 2010; Abosti *et al.*, 2012).

The events involved in the vascular response to carrageenan-induced oedema are similar to the early exudative stage of inflammation (Winter *et al.*, 1962; Ozaki, 1990) and hence the use of anti-inflammatory agent to inhibit this acute phase of inflammation mitigates the inflammatory process. However, it must be noted that, anti-inflammatory activity of a drug or an agent administered prophylactically does not certainly imply its ability to work therapeutically. For instance, cyclosporine prevented the inception of collagen-induced inflammation in rats but could not treat it after the inception of inflammation (Kaibara *et al.*, 1983).

HLML, LOML and HLMR significantly (p < 0.001) at 300 mg/kg reduced the induced oedema during both prophylaxis and therapeutic treatments. This indicates the presence of compounds that can reduce the inflammatory responses. Anti-

inflammatory activity of ethanol leaf extract of *H. latifolia* has also been reported by Abosti *et al.* (2012), using the carrageenan induced oedema in 7-day old chicks model. In addition to the leaves of *H. latifolia*, the anti-inflammatory activity of the root was also determined using the carrageenan induced oedema in rats. The root extract (HLMR) also exhibited anti-inflammatory activity by significantly (p<0.001) reducing induced oedema at 300 and 100 mg/kg in both prophylactic and therapeutic experiments.

Even though the mechanisms of action of HLML, LOML and HLMR are not known, they could be acting by inhibition and/or interference in the action of inflammatory mediators (such as histamine, serotonin, bradykinin, prostaglandins and other cyclooxygenase products) involved in the carrageenan induced oedema (Abotsi *et al.*, 2012; Obiri *et al.*, 2013).

Also phytochemicals such as tannins, glycosides, sterols, terpenoids, flavonoids and alkaloids present in HLML, LOML and HLMR may contribute to their anti-

inflammatory activities. Some of these secondary metabolites present in plant extracts have been reported to exhibit anti-inflammatory properties (Guardia *et al.*, 2001; Rotelli *et al.*, 2003; Barbosa-Filho *et al.*, 2006).

Furthermore, the mechanism of anti-inflammatory action of HLML, LOML and HLMR may also be due to their antioxidant properties. ROS are released from activated neutrophils and macrophages during inflammatory injury and their overproduction leads to tissue injury or damage. ROS can also cause the release of pro-inflammatory cytokines such as IL-1 and TNF- $\alpha$  which directly enhance inflammatory response and also stimulate the production of neutrophils and macrophages (Dinarello, 2000; Conforti *et al.*, 2008). Hence, the extracts ability to neutralize the effects of ROS and free radicals can contribute to their antiinflammatory activities.

The ethnomedicinal use of *H. latifolia* and *L. ovalifolia* as wound healing agents (Agyare *et al.*, 2009) was investigated using the excision wound model in rats. Wounds treated with 5 and 10% HLML, HLMR and LOML formulated aqueous creams enhanced high rate of wound contraction as indicated by centripetal movement of the edges of a full-thickness wound (Suguna *et al.*, 2002, Tang *et al.*, 2007). Rate of wound contraction in excised wounds treated with 5 and 10% w/w of HLML, HLMR and LOML significantly (p<0.001) increased from day five (5) up to day fifteen (15) when compared with the untreated group (Figures 3.8.1 to 3.8.3).

Again, histological investigations of the treated wounds revealed high fibroblast proliferation, angiogenesis and granulation, collagenation, tissue formation and reepithelisation in treated wound tissues compared to the untreated wound tissues. Wound contraction is indicated by enhanced re-epithelialisation, granulation, angiogenesis, fibroblast proliferation, keratinocyte differentiation and proliferation (Tang *et al.*, 2007).

The wound healing activity of HLML, HLMR and LOML may be attributed to their antioxidant and anti-inflammatory properties. Over-production of ROS and prolonged inflammation can result in impaired wound healing (Martin and Leibovich, 2005; Shetty *et al.*, 2008). Therefore an agent ability to mop up excess ROS or free radicals can minimize the inflammatory response at wound site which can also enhance the wound healing process.

During development of human skin towards an intact barrier system, keratinocytes will undergo cellular proliferation followed by a switch to cellular differentiation to restore the breakage in the intact skin (Agyare *et al.*, 2011). HLML (p<0.001) and HLMR (p<0.01) at 100 µg/mL significantly reduced the viability of HaCaTkeratinocytes compared to the untreated cells. This may suggest that HLML and HLMR may be cytotoxic at higher concentrations. Also, HLML, HLMR and LOML, though not significant (p>0.05), inhibited the release of LDH from the cytoplasm of HaCaTkeratinocytes compared to the untreated cells. Low LDH release from the cytoplasm by the extracts suggests that HLML, HLMR and LOML may not be cytotoxic agents at the concentrations used. There is a need to isolate and identify the bioactive compounds in the extracts responsible for the above biological activities.

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#### **CHAPTER FIVE**

## 5.0 CONCLUSION AND RECOMMENDATIONS

### **5.1 CONCLUSION**

The leaf and root extracts of *H. latifolia* and the leaf extract of *L. ovalifolia* exhibited antimicrobial, antioxidant, anti-inflammatory and wound healing activities. HLML, HLMR and LOML showed a broad spectrum antimicrobial activity against test organisms at high concentrations. The sub-inhibitory concentration of the extracts also modified the activity of some antibiotics by either enhancing or reducing their antimicrobial activity. HLML, HLMR and LOML exhibited both prophylactic and therapeutic acute anti-inflammatory activities. The extracts did not show cytotoxic effect on HaCaT-keratinocytes at the concentration used.

# 5.2 RECOMMENDATIONS

- The mechanism of action of the extracts for anti-inflammatory activity should be investigated.
- *In vivo* toxicity studies should be conducted on the plants to establish their toxicity potentials in living tissues.
- Active compounds responsible for anti-inflammatory and wound healing properties should be isolated and characterized.

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#### APPENDICE

#### **APPENDIX 1: Composition and preparation of microbial media**

#### A1.1 Nutrient agar (Oxoid, Basingstoke, UK)

Table 1.1: Composition of Nutrient agar

Composition	Quantity (g)	ICT
Lab-lemco powder	1.0	
Yeast extract	2.0	
Peptone	5.0	
Sodium chloride	5.0	
Agar	15.0	

A quantity of 28.0 g Nutrient agar powder was weighed into a beaker. Eight hundred millilitres of sterile distilled water was added and heated at 100°C until it begins to froth. The resulting mixture was then made up to 1 L with sterile distilled water and transferred in 20 mL quantities into glass test tubes. The mixture was then sterilised in an autoclave at 121°C for 15 minutes.

#### A1.2 Nutrient broth (Oxoid, Basingstoke, UK)

Table 1.2: Composition of Nutrient broth

Composition	Quantity (g)	F
Lab-lemco powder	1.0	-
Yeast extract	2.0	2
Peptone	5.0	
Sodium chloride	5.0	
dium chloride	5.0	

A quantity of 13.0 g Nutrient broth powder was weighed into a beaker. Four hundred millilitres of sterile distilled water was added and stirred to dissolve. Sterile distilled water was added to prouce 1 L solution. Ten millilitres of the solution was transferred into glass test tubes and sterilized at 121°C for 15 minutes in an autoclave.

#### A1.3 MacConkey agar (Oxoid, Basingstoke, UK)

Table 1.3: Composition of MacCo	onkey agar
Composition	Quantity (g)
Peptone	20.00
Lactose	10.00
Bile salts	1.50
Sodium chloride	5.00
Neutral red	0.03
Crystal violet	0.001

A quantity of 51.5 g MacConkey agar powder was weighed into a beaker and suspended in 800 mL sterile distilled water. The mixture was boiled, made up to a volume of 1 L and sterilized in an autoclave at 121°C for 15 minutes.

# A1.4 Mannitol salt agar (Oxoid, Basingstoke, UK)

Table 1.4: Composition of	Mannitol salt agar	IICT
Composition	Quantity (g)	
Lab-lemco powder	1.00	
Peptone	10.00	
Mannitol	10.00	
Sodium chloride	75.00	
Phenol Red	0.025	
Agar	15.0	1 141

A quantity of 111.0 g Mannitol salt agar powder was weighed into a beaker and suspended in 800 mL of sterile distilled water. The mixture was boiled, made up to 1 L with distill water and sterilized in an autoclave at 121°C for 15 min.

#### A1.5 Cetrimide agar (Oxoid, Basingstoke, UK)

 Table 1.5: Composition of Cetrimide agar

Composition	Quantity (g)	
Gelatin peptone	20.00	
Magnesium chloride	1.40	
Potassium sulphate	10.00	
Cetrimide	0.30	
Agar	13.60	

A quantity of 45.3 g Cetrimide agar powder was weighed into a beaker and suspended in 800 mL sterile distilled water. The mixture was boiled, made up to 1 L and sterilized in an autoclave at 121°C for 15 min.

# A1.6 Bismuth sulphite agar (Oxoid, Basingstoke, UK)

Table 1.6: Composition of Bismu	uth sulphite agar
Composition	Quantity (g)
Peptone	5.00
Lab-lemco powder	5.00
Glucose	5.00
Di-sodium phosphate	4.00
Ferrous sulphate	0.30

Bismuth sulphite indicator	8.00
Brilliant green	0.02
Agar	12.70

A quantity of 40 g of Bismuth sulphite agar powder was weighed into a beaker and suspended in 800 mL sterile distilled water. The mixture was heated until it began to boil and simmer. The suspension was made up to 1 L whiles heating. The suspension was then transfered into glass test tubes.

#### A1.7 Potato dextrose agar (Oxoid, Basingstoke, UK)

Table 1.7: Composition of Potato	dextrose agar
Composition	Quantity (g)
Potato extract	4.00
Dextrose	20.00
Agar	15.00

A quantity of 39.0 g of potato dextrose agar powder was weighed into a beaker and suspended to 800 mL sterile distilled water. The mixture was boiled, made up to 1 L and sterilized in an autoclave at 121°C for 15 min.

#### A1.8 Koser's citrate medium (Oxoid, Basingstoke, UK)

Table 1.8: Composition of Koser's of	citrate medium
Composition	Quantity (g)
Sodium Ammonium Phosphate	1.50
Potassium Dihydrogen Phosphate	1.00
Magnesium sulphate	0.20
Sodium citrate	2.50
Bromothymol blue	0.016

A quantity of 5.2 g of the powder was weighed into a beaker containing 800 mL of sterile distilled water and stirred to dissolve. Sterile distilled water was added to produce a final volume of 1 L. Ten (10) millilitres of the solution was then distributed into glass test tubes and sterilized at 121°C for 15 minutes in an autoclave.

#### A1.9 Tryptone broth (Oxoid, Basingstoke, UK)

Table1. 9: Composition of Potato	dextrose agar
Composition	Quantity (g)
Tryptone	10.00
Sodium chloride	5.00

A quantity of 15.0 g of the powder was weighed into a beaker and dissolved in 800 mL of sterile distilled water. Enough sterile distilled water was added to produce a final voume of 1 L. Ten (10) millilitres of the mixture was then distributed into glass test tubes and sterilized at 121°C for 15 minutes in an autoclave.

#### A1.10 MRVP medium (Oxoid, Basingstoke, UK)

Table 1.10: Composition of MRV	P medium
Composition	Quantity (g)
Peptone	7.00
Glucose	5.00
Phosphate buffer	5.00

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A quantity of 17.0 g of the powder was weighed into a beaker and dissolved in 800 mL of sterile distilled water. Enough sterile distilled water was added to produce a final volume of 1 L. Ten (10) millilitres of the mixture was then distributed into glass test tubes and sterilized at 121°C for 15 minutes in an autoclave.



### **APPENDIX 2:** Confirmatory tests on test organisms

Confirmatory test was performed on both the typed and clinical strains used in the study to confirm their identity.

#### A2.1 Staphylococcus aureus

*Staphylococcus aureus* was identified by culturing a suspension of organisms on mannitol salt agar followed by coagulsae test. The appearance of yellow colonies on the surface of the agar after 24 h incubation at 37°C indicates the presence of *S. aureus*. Furthermore, the isolated organisms on mannitol salt agar are re-cultured in nutrient broth. A quantity of 0.1 mL of a 24 h broth culture of the isolated colonies was inoculated into nutrient agar containing 10% v/v rabbit blood plasma and incubated for 24 h at 37°C. The coagulation of the plasma (coagulase positive) confirms the suspension to be *S. aureus*.

# A2.2 Bacillus subtilis

Starch and casein hydrolysis was used to identify *B. subtilis*. Nutrient agar containing 10% w/v starch was prepared. A loopful of 24 h broth culture of the suspension of organism was streaked at the middle on the surface of nutrient agar containing 10% w/v starch and incubated at 37°C for 24 h. After incubation the plate was then sprayed with iodine solution. Clear region due to the hydrolysis of starch were seen against a blue black surrounding. The clear zones were indicative of the presence of *B. subtilis*.

#### A2.3 Streptococcus pyogenes

The identification of S. pyogenes was performed with blood agar. Nutrient agar was melted and mixed with about 5% defribinated horse blood and poured into a petri dish. A 24 h broth culture of the organism was placed on the surface of the agar using a straightened loop as spots and incubated at 37 °C for 24 h. The presence of clear zones around the organism spots made indicates the haemolysis of red blood cells ( $\beta$ -haemolysis) hence the presence of S. pyogenes.

#### A2.4 Escherichia coli

MacConkey agar was seeded with the organisms by streaking and the plate incubated at 37°C for 24 h. The appearance of red-violet colonies on the surface of the agar after incubation indicates the presence of *E. coli*. The indole and Methy Red-Voges Proskauer (MRVP) tests were used as confirmatory test where this organism is known to be MR positive and VP negative. In the indole test 0.1 mL of the suspension of organism is inoculated in tryptone broth and incubated at 37°C for 24 h. After incubation Kovac's reagent was added to the tryptone broth. The appearance of a pink/red ring at the interface or meniscus of the broth indicated the presence of *E. coli*. The MRVP test was performed by inoculating a 24 h broth culture of the organisms in MRVP broth.

After incubation the culture was divided into two for the distinct MR and VP tests. In the MR test methyl red was added to the culture and the appearance of pink/red colouration shows that the organism is MR positive.  $\alpha$ -naphthol and KOH solutions

were added to the second part of the broth culture for the VP test. The absence of a cherry red colouration indicates that the organism was VP negative.

# A2.5 Pseudomonas aeruginosa

Two loopfuls of a suspension of the microorganisms was streaked on cetrimide agar and incubated at 37°C for 24 h incubation. The observation of greenish colonies on the surface of the agar after incubation indicates the presence of *P. aeruginosa*. Ten (10) milliliters of Koser's citrate medium was inoculated with a loopful of the colonies. The conversion of the green Koser's citrate to blue colouration confirmed the identity of *P. aeruginosa*.

#### A2.6 Salmonella typhi

Bismuth sulphite agar was inoculated with two loopfuls of the microorganisms and incubated at 37°C for 24 h. Black colonies surrounded by metallic sheen observed within 18-24 hours of incubation indicates the presence of *S. typhi*.

#### A2.7 Klebsiella pneumoniae

Indole, MRVP and catalase tests were used to confirm *K. pneumoniae*. The indole test was performed by inoculating 0.1mL of a 24 h broth culture of the organism in typtone broth and incubating at 37 °C for 24 h. The absence of a pink/red colour after the addition of Kovac's reagent to the overnight culture indicates the presence of *K. pneumoniae*. MRVP test was performed by inoculating a 24 h broth culture of the organisms in MRVP broth. The culture after incubation was divided into two; one part for the MR test and the other for the VP test. The MR test was performed by the addition of methyl red to the culture; the appearance of yellow colouration indicates that the organism is MR negative. The VP test was performed by the addition of  $\alpha$ -naphthol and KOH solutions; the appearance of a cherry red colouration indicates that the organism was VP positive. The catalase test was performed by placing 1 mL of fresh 3 % hydrogen peroxide into a glass test tube and inoculating with about 0.5mL of the bacteria culture. The appearance of air bubbles indicates the presence of catalase enzyme in *K. pneumonia*.

#### A2.8 Candida albicans

A 24 h broth culture of the suspension of organisms was streaked on the surface of potato dextrose agar containing 1 g chloramphenicol and incubated at  $25^{\circ}$ C for 72 h. Two loopfuls of colonies were streaked on the surface of eosin methylene blue agar. The appearance of white colonies indicated the presence of *C. albicans*.

# **APPENDIX 3: Effect of sub-inhibitory concentration (5mg/mL) of HLML and LOML on some antibiotics**

Table 3.1: MICs of antibiotics only and antibiotics in the presence of sub-inhibitory concentration of extracts

Antibiotic	s only/	EC	BS	ST	KP	SA	EF	SP	PA
antibiotics HLML/LC	s + DML				MIC	C (µg/mL)	)		
Amoyy	Only	256	128	128	128	256	>1024	256	128
	HLML	32	32	64	128	32	>1024	>1024	64
	LOML	128	128	256	256	128	>1024	>1024	128
	Only	128	32	128	128	64	>1024	64	64
Ampı	HLML	64	32	32	128	64	>1024	>1024	64
	LOML	128	128	128	256	64	>1024	>1024	>1024
	Only	16	32	32	64	32	32	32	64
Tetra	HLML	32	32	32	16	32	32	32	32
ç	LOML	128	<mark>256</mark>	128	128	128	128	128	128
	Only	32	256	64	128	64	128	64	128
Erythro	HLML	512	1024	1024	1024	1024	1024	512	512
	LOML	512	512	512	1024	512	512	1024	1024
C.	Only	2	2	4	4	4	4	4	4
Cipro	HLML	16	16	16	8	8	8	8	16
	LOML	16	32	32	32	32	16	16	32
		×	W	SAI	NE 1	0	A		

# **APPENDIX 4: Antioxidant activity**

-	-			- 0 0	,						-	
conc		Vit E			HLML			HLMR	2		LOML	4
(µg/mL)												
15.6	21.9	21.3	21.8	6.5	6.5	6.6	1.6	1.1	1.7	5.8	6.5	5.8
31.3	30.6	30.5	30.4	10.9	10.9	10.4	2.6	3.5	2.7	9.5	10.9	9.5
62.5	56.5	54.9	56.1	21.4	21.4	22.3	9.7	9.8	9.3	17.3	18.5	16.3
125.0	77.8	78.7	78.2	29.2	29.4	30.7	13.7	13.5	16.2	24.8	24.3	24.8
250.0	78.7	78.7	75.5	35.8	37.7	37.3	20.4	20.1	18.5	29.4	27.2	29.4
500.0	76.5	78.0	78.5	36.5	40.7	38.3	21.2	23.3	23.6	37.9	35.5	37.9
1000.0	75.3	79.3	73.0	50.6	50.8	53.8	33.2	33.2	34.7	42.3	41.0	42.3

A4.1 DPPH free radical scavenging activity Table 3.1: Percentage inhibition



Figure 4.2: Tannic acid calibration curve for total phenol content determination

	-				-				
Conc	1	LML	-		LOML		1	HLMR	
(mg/mL)	1	90				-	0	/	
10.0	194.91	200.55	198.50	120.06	119.96	124.89	166.57	174.17	186.69
8.0	187.93	191.21	187.21	101.79	101.48	99.73	155.17	154.46	155.07
6.0	167.19	171.09	157.23	80.12	80.84	80.64	100.04	109.90	92.24
4.0	92.85	95.63	106.51	54.66	53.02	55.17	79.30	96.34	87.21
2.0	26.32	34.33	37.93	21.60	15.65	18.73	42.75	75.20	63.29
1.0	24.37	27.04	31.77	13.59	14.31	14.11	21.60	37.10	39.77
0.5	8.46	9.79	11.54	7.84	7.54	6.10	14.41	26.53	39.88

	<b>Table 4.2: To</b>	otal phenol	content of	extracts ex	pressed as	tannic acid e	quivalence
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A4.3 Total antioxidant capacity

L



**Figure 4.3:** α-tocopherol calibration curve for total antioxidant capacity determination

			e enparente	,	tets empire		· · · · · · · · · · · · · · · · · · ·		
Conc		HLML	1	~	LOML	7		HLMR	
(mg/mL)				-	_				
10.0	705.24	715.42	676.95	447.46	488.46	501.18	558.39	645.17	662.97
8.0	532.33	509.76	509.44	473.84	450.00	470.34	437.60	545.04	546.63
6.0	488.14	528.19	512.62	427.11	431.88	395.65	431.56	479.56	539.32
4.0	453.18	384.84	385.47	332.07	395.96	380.39	379.12	379.75	422.66
2.0	276.45	326.03	288.21	216.69	258.33	300.29	268.50	321.90	326.99
1.0	243.07	229.08	241.80	192.53	196.03	199.21	248.47	<u>307.60</u>	269.45
0.5	240.21	150.57	220.50	177.59	173.78	179.82	273.59	271.68	251.65

Table 4.3: Total antioxidant	canacity of	extracts expressed	as $\alpha$ -tocophe	rol equivalence
	capacity of	CALLACTO CAPICOSCU	as u-tocopie	u u u u u u u u u u

#### **APPENDIX 5:** Anti-inflammatory activity

#### A 5.1 Preparation of Carrageenan suspension

A 1 % w/v carrageenan suspension was prepared by suspending 100 mg of carrageenan powder in 10 mL of 0.9 % w/v saline solution. The suspension was stirred to ensure uniform suspension and allowed to stand on the bench at  $25^{\circ}$ C for 23 hours before use.

#### A 5.2 Prophylactic anti-inflammatory activity of extracts.

time/hr	0	1	2	3	4	5	6
	0,00,0,00	56 58	62.35	56 79	52.67	46 30	45.88
Control	0.00 0.00	57.87	62.58	62 17	56.85	53 58	49.69
Control	0.00	55.65	57.20	52.17	48.25	12 22	40.04
	0.00	55.05	51.29	55.10	46.23	45.55	40.04
	0.00	62.76	63.81	60.88	55.02	53.97	52.72
			1 / 15		1.0	and the second second	
	$0.00\ 0.00$	47.16	41.29	29.55	24.43	16.48	13.07
Aspirin	0.00	46.12	42.04	40.20	29.39	21.84	16.94
100 mg/mL		37.66	32.98	30.21	29.36	25.11	18.94
C	0.00	40.29	33.68	29.13	26.86	24.17	23.35
	0.00.0.00	50.97	53.98	52.26	45.81	41 72	36.56
LOMI	0.00 0.00	52 39	55.13	53 53	48.97	46 70	<i>44</i> 19
20  mg/l/s	0.00	53.15	55 75	42 52	47.94	36.01	34.06
50 mg/kg	0.00	51.75	55.15	54.00	12.02	20.10	25.11
	0.00	51.75	56.67	54.00	42.92	38.19	35.11
	0.00 0.00	47.69	50.00	46.22	40.97	35.50	31.30
LOML	0.00	43.72	45.14	39.68	35.43	32.79	28.34
100 mg/kg		48.89	55.35	41.01	38.38	34.34	29.49
6 6	0.00	49.69	54.09	51.36	39.20	35.01	32.08
	0.00	.,,,,,,	0.1103	01100	07.20	00101	02.00
9	0.00 0.00	41.43	41.83	39.04	32.47	27.89	22.71
LOML	0.00	38.63	39.02	36.47	32.94	29.22	22.55
300 mg/kg		45.07	48.09	39.03	36.02	31.79	24.95
	0.00	45.97	51.27	49.15	36.86	30.93	29.87
	0.00 0.00	40.31	45.49	41.65	29.37	27.64	26.30
HI MI	0.00	37.90	48.41	47.84	46.15	40.34	37.90
30  mg/kg		35.10	38.71	37.19	31.12	22.39	20.49
50 mg/kg	0.00	28 21	34.63	32.68	31.13	28.21	24.51
	0.00	20.21	54.05	52.00	51.15	20.21	24.51
	0.00 0.00	47.00	58.59	55.28	49.69	44.10	42.24
HLML	0.00	43.91	55.51	51.84	47.00	44.10	41.59
100 mg/kg		26.12	3 <mark>3.</mark> 96	31.34	26.12	19.03	17.72
	0.00	55.68	56.80	<mark>56.4</mark> 2	<mark>55</mark> .87	53.82	47.4 <mark>9</mark>
	0.00.0.00	26.73	41 31	40.00	34 39	25.05	20.75
ні мі	0.00	26.98	39.81	34 72	28.68	24.53	23.77
300  mg/kg	0.00	31.10	42.72	36.81	35.63	32.28	31.10
JOU IIIg/Kg	0.00	20.94	28 69	27.24	24.67	21.02	17.07
	0.00	20.84	20.08	27.34	24.07	21.05	1/.9/
HLMR	0.00 0.00	50.38	54.18	52.66	50.57	46.39	45.06
30 mg/kg		41.10	52.12	48.94	47.88	43.22	41.53
	0.00	51.59	57.94	52.38	45.63	44.64	41.67
	0.00	50.00	51.91	49.62	48.47	46.37	42.18

 Table 5.2.1 Percentage paw eodema inhibition of aspirin, negative control LOML,

 HLML and HLMR extracts in prophylaxis studies

HLMR	0.00 0.00 0.00	48.05 40.96	49.13 47.69	40.69 40.96	37.66 37.12	32.90 32.50	31.39 27.50
100 mg/kg	0.00	45.68 43.29	47.16 49 34	42.53 44.05	37.47 40.08	35.16 34 97	32.42 31.57
	0.00	-13.27	-7.5-	44.05	40.00	54.77	51.57
HLMR 300 mg/kg	0.00 0.00 0.00	45.74 29.50 35.22	47.52 36.18 39.92	39.41 34.32 34.46	36.44 30.43 29.38	30.50 27.27 22.03	25.54 25.97 19.40
	0.00	35.96	41.57	36.89	35.77	32.40	26.03
			$\mathbf{N}$	ΛL	73		

# A 5.3 Therapeutic anti-inflammatory activity of extracts.

Table 5.3.1 Percentage paw eodema inhibition of aspirin, negative control LOML,HLML and HLMR extracts in therapeutic studies

time/hr	0	1	2	3	4	5 6		7
Control	0.00	28.91	89.94	103.98	116.51	113.09 110.44		102.09
	0.00	28.91	69.47	88.00	96.00	90.53	88.42	85.89
				100				
	0.00	28.91	67.81	91.42	100.00	96.57	94.85	90.99
100	0.00	28.91	93.83	101 65	1 2	98 77	95 47	91 77
	0.00	20.71	25.05	34 29	108.23	20.11	JJ.+7	)1.//
Aspirin	0.00	28.91	33.06	38.12	29.39	27.55 35.33	23.06	13.06
100	0.00	28.91	37.33	38.11	37.52	34.18	34.13	27.74
mg/kg	0.00	28.91	31.63	00111	36.94		29.86	24.56
00	0.00	28.91	28.91	38.22	34.46	31.68	29.31	23.17
			-	200	× ×		2	
LOML	0.00	28.91	38.80	41.40	37.80	<u>30.80 34.88</u>	26.00	21.60
300	0.00	28.91	39.72	45.77	40.93	30.75	30.04	28.43
mg/kg	0.00	28.91	38.09	42.77	37.88		30.14	23.22
	0.00	28.91	40.08	47.35	36.15	32.02	<mark>30</mark> .45	25.54
					2.5	11		( · · · · ·
LOML	0.00	28.91	47.60	50.52	46.14	41.54 39.02	35.07	27.14
100	0.00	28.91	45.49	44.71	42.35	33.70	36.27	30.59
mg/kg	0.00	28.91	39.93	45 <mark>.6</mark> 0	40.66		31.68	28.39
	0.00	28.91	45.56	<u>48.52</u>	46.48	34.81	33.33	28.15
LONG	0.00	20.01	10.05	52.50	17.10	40.15	2614	24.61
LOML	0.00	28.91	48.95	52.58	47.42	40.15	36.14	34.61
30  mg/kg	0.00	28.91	42.88	45.29	44.18	40.67	35.67	34.01
	0.00	28.91	43.22	44.57	45.16	42.05	39.15	32.95
	0.00	28.01	52 11	40.10	45 20	44.00	27 00	22.27
	0.00	20.91	33.11	49.10	45.29	44.09	57.00	55.27
HLML	0.00	28.91	54.45	62.30	61.78	56.20 63.33	47.64	44.33
300	0.00	28.91	68.25	74.91	70.00	68.64	57.72	54.04
mg/kg	0.00	28.91	71.60	74.39	73.34		58.71	56.27
00	0.00	28.91	65.93	76.35	73.35	66.73	64.53	63.13

HLML	0.00	28.91	45.32	57.30	55.81	51.69 48.89	48.50	45.32
100	0.00	28.91	50.93	51.85	55.19	55.77	43.70	35.37
mg/kg	0.00	28.91	63.08	71.54	62.50		48.08	45.77
00	0.00	28.91	60.97	65.47	61.15	58.99	52.16	48.38
HLML	0.00	28.91	66.78	72.03	66.44	64.24 75.18	61.02	59.83
30 mg/kg	0.00	28.91	69.64	82.14	80.71		74.11	73.04
	0.00	28.91	43.83	51.67	47.67	41.33	38.83	36.83
				1.201	ъ. т.	1.1.00	-	
	0.00	28.91	51.25	61.21	53.02	50.36	48.75	41.64
				K				
HLMR	0.00	28.91	52.60	56.08	56.77	53.82 42.64	52.26	51.74
300	0.00	28.91	48.05	44.50	60.74	64.15	32.99	31.98
mg/kg	0.00	28.91	48.58	59.13	65.83		61.64	54.10
	0.00	28.91	64.63	73.17	79.44	72.47	70.56	56.27
HLMR	0.00	28.91	46.38	51.68	53.44	51.50 83.05	49.03	40.04
100	0.00	28.91	77.90	88.76	86.67	69.87	79.81	69.71
mg/kg	0.00	28.91	71.72	77.95	78.45	1 1 1 1	62.12	57.24
	0.00	28.91	68.05	72.74	82.33	76.69	74.06	69.74
				-				
HLMR	0.00	28.91	68.92	76.23	75.50	66.91 60.47	63.44	57.22
30 mg/kg	0.00	28.91	60.29	71.20	69.77		57.42	52.24
	0.00	28.91	57.99	65.99	58.36	53.72	51.30	46.84
	0.00	28.91	51.89	55.83	51.17	47.04	42.37	38.42
		-				1-2-	-	-

# **APPENDIX 6: Wound healing activity A6.1 Preparation of aqueous cream**

The aqueous cream used for the wound healing experiment was prepared according to the method described in the British Pharmacopoeia (BP) 2000. The preservative was not included in the preparation to prevent its interference with the antimicrobial activity of the extracts.

# Table 6.1.1: Preparation of emulsifying ointment

EMULSIFYING OINTMENT								
A A A A	Master formula (g) Ingredients							
Emulsifying wax White soft paraffin	300 500							
Liquid paraffin	200							
Total	1000							

1	AQUEOUS CREAM BP
Ingredients	Master formula (g)
Emulsifying ointment	300
Phenoxyethanol	10
Sterile distilled water	690

# Table 6.1.2: Preparation of aqueous cream BP AOUFOUS CREAM BP

The phenoxyethanol was excluded during the preparation and replaced with distilled water. The weight of water was calculated from its density (1 g/mL).

One hundred grams of aqueous cream was prepared by mixing 30 g emulsifying ointment in 70 mL of sterile distilled water. The mixture was heated on a water bath and stirred in one direction until a homogenous mixture was obtained. The mixture was then allowed to cool. The extracts were incorporated into the homogenous cream to formulate creams of various concentrations of extracts (5 and 10% w/w).

LONIL-II									
Days	0	4	3	5	7	9	11	13	15
Untreated	0.00	26	15.79	21.05	26.32	31.58	42.11	47.37	57.89
	0.00	0.00	5.00	10.00	15.00	20.00	30.00	45.00	55.00
	0.00	4.76	9.52	9.52	14.29	19.05	23.81	33.33	42.86
	0.00	5.00	10.00	15.00	20.00	25.00	25.00	35.00	45.00
	0.00	5.26 5.88	10.53	15.79	21.05	26.32	31.58	36.84	47.37
Vehicle	0.00	10.53	17.65	23.53	29.41	35.29	41.18	47.06	52.94
	0.00	5.88	21.05	21.05	26.32	31.58	42.11	47.37	57.89
	0.00		11.76	17.65	23.53	35.29	47.06	52.94	52.94
	0.00	11.11	11.11	27.78	33.33	33.33	44.44	44.44	50.00
	0.00	-	10.53	15.79	21.05	26.32	<mark>36.8</mark> 4	42.11	47.37
	0.00	5.26	20.00	32.00	44.00	48.00	56.00	64.00	80.00
	0.00	12.00	16.67	25.00	37.50	41.67	54.17	66.67	83.33
SS 1%w/w	0.00	12.50 8.33	12.50	20.83	29.17	37.50	50.00	66.67	79.17
	0.00	7.41	14.81	25.93	40.74	55.56	62.96	74.07	81.48
	0.00	9.09 8.70	22.73	31.82	40.91	54.55	63.64	72.73	81.82
10% HLML	0.00	8.00	21.74	30.43	34.78	39.13	52.17	60.87	78.26
	0.00	9.09	24.00	32.00	40.00	52.00	56.00	68.00	76.00
	0.00		13.64	27.27	36.36	45.45	54.55	68.18	77.27
	0.00	9.52	19.05	28.57	38.10	47.62	61.90	71.43	80.95

 Table 6.1.2: Percentage wound closure of untreated, vehicle, HLML, HLMR and LOML-treated wounds

	0.00	12.00	20.00	28.00	36.00	48.00	60.00	72.00	80.00
	0.00	8.33	16.67	29.17	37.50	37.50	50.00	58.33	75.00
	0.00	13.64	22.73	31.82	36.36	36.36	45.45	59.09	77.27
5% HLML	0.00	14.29	23.81	33.33	42.86	47.62	57.14	66.67	76.19
570 HENTE	0.00	8.00	16.00	28.00	40.00	48.00	56.00	64.00	72.00
	0.00	12.50	20.83	33.33	45.83	54.17	62.50	70.83	75.00
10% HLMR	0.00	17.39	26.09	34.78	47.83	47.83	56.52	65.22	82.61
	0.00	13.64	27.27	31.82	36.36	50.00	59.09	68.18	81.82
	0.00	8.33	20.83	33.33	45.83	54.17	62.50	70.83	79.17
	0.00	13.04	21.74	30.43	39.13	56.52	65.22	73.91	82.61
	0.00		16.00	24.00	36.00	48.00	56.00	64.00	80.00
	0.00	8.00	24.00	32.00	44.00	48.00	52.00	64.00	76.00
	0.00	12.00	26.09	34.78	43.48	47.83	56.52	65.22	78.26
5% HLMR	0.00	17.39	18.18	27.27	40.91	50.00	59.09	68.18	77.27
	0.00	9.09	20.00	28.00	40.00	48.00	56.00	64.00	72.00
	0.00	8.00 8.70	- 17.39	26.09	30.43	39.13	47.83	56.52	73.91
		8.00	10						
10% LOML	0.00	12.50	20.00	32.00	44.00	48.00	52.00	64.00	84.00
	0.00	13.04	25.00	37.50	45.83	50.00	58.33	70.83	79.17
	0.00		26.09	39.13	47.83	56.52	65.22	73.91	82.61
	0.00	8.00	16.00	28.00	40.00	56.00	64.00	72.00	80.00
	0.00	4.35 <sup>0</sup>	- 17.39	30 <mark>.4</mark> 3	39.13	52.17	65.22	73.91	82.61
5% LOML	0.00	8.7	17.39	26.09	34.78	34.78	52.17	60.87	78.26
	0.00	8.33	16.67	29.17	41.67	45.83	58.33	66.67	75.00
	0.00	9.09	18.18	27.27	31.82	36.36	45.45	54.55	72.73
	0.00	9.52	19.05	<mark>28.</mark> 57	38.10	47.62	57.14	66.67	76.19
	0.00	12.50	20.83	25.00	33.33	41.67	50.00	62.50	70.83

